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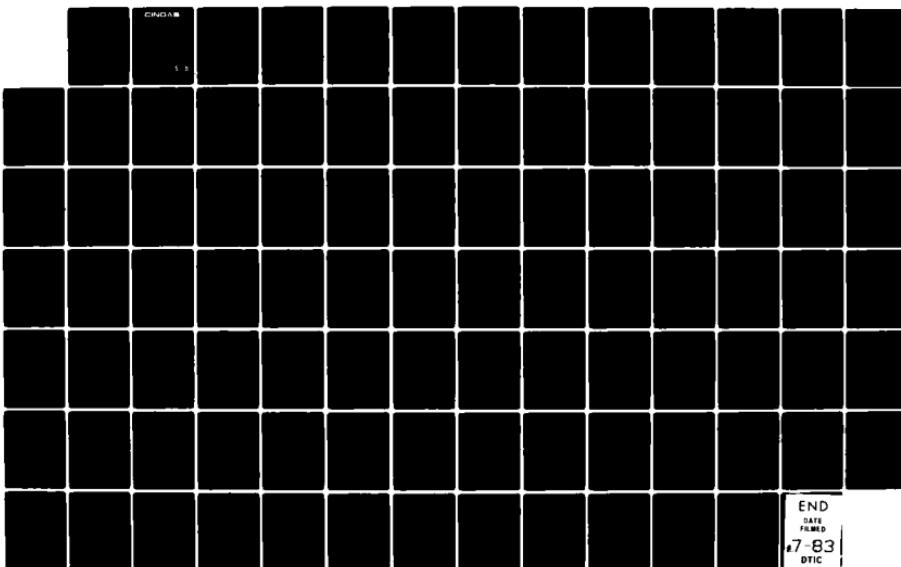
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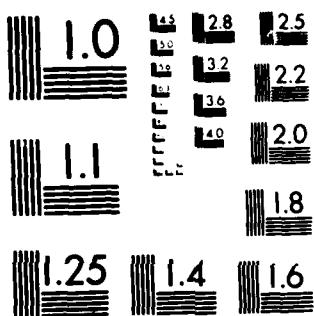
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ELECTRICAL RESISTIVITY OF ALUMINUM AND MANGANESE

By

P. D. Desai, H. M. James, and C. Y. Ho

CINDAS Report 65

March 1983

Prepared for

OFFICE OF STANDARD REFERENCE DATA
National Bureau of Standards
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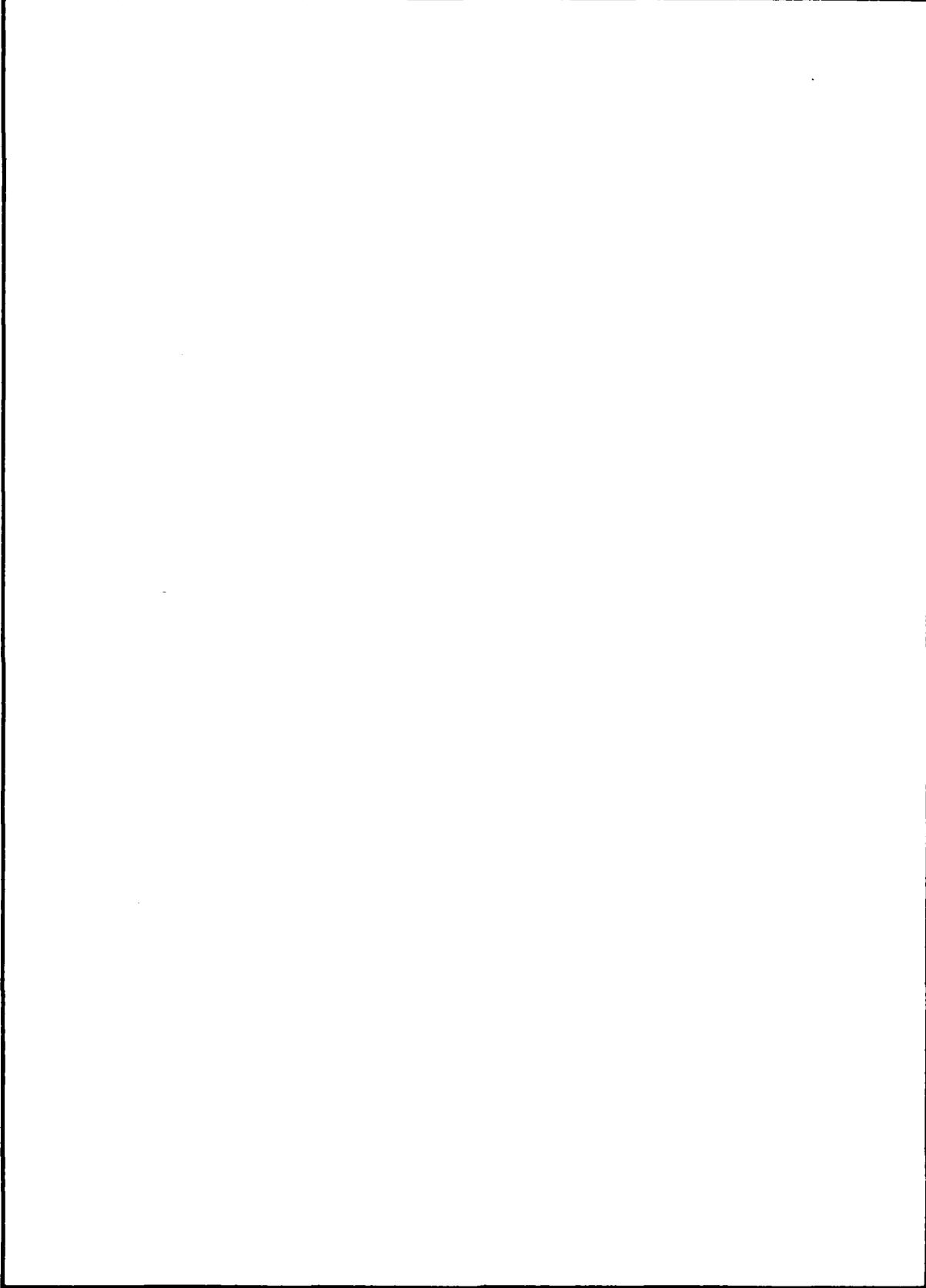
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PREFACE

This technical report was prepared by the Center for Information and Numerical Data Analysis and Synthesis (CINDAS), Purdue University, West Lafayette, Indiana, under the auspices of the Office of Standard Reference Data of the National Bureau of Standards (NBS), Department of Commerce, Washington, D.C.

This report represents the most exhaustive compilation and critical evaluation of the recorded world knowledge on the electrical resistivity of aluminum and manganese and is one of a series of technical reports on the electrical resistivity of selected elements. The literature search and data compilation have been done in a most extensive and detailed manner, making it possible for all users of the subject to have access to the original data without having to duplicate the laborious and costly process of literature search and data extraction. Also, for the active researchers in the field, a detailed discussion is presented for each material, reviewing the available data and information, giving details of data analysis and synthesis, and discussing the considerations involved in arriving at the final recommended values.

It is hoped that this work will prove useful not only to the engineers and scientists in the field but also to other engineering research and development programs and for industrial applications, as it provides a wealth of knowledge heretofore unknown or inaccessible to many. In particular, it is thought that the critical evaluation, analysis and synthesis, and reference data generation constitute a unique aspect of this work.

Although this report is primarily the result of financial support and interest of the NBS Office of Standard Reference Data, the extensive documentary activity essential to this work was supported by the Defense Logistics Agency of the Department of Defense. Thanks are due Dr. H. J. White, Jr., of the NBS Office of Standard Reference Data for his guidance, cooperation, and sympathetic understanding during the course of this work.

ABSTRACT

This work compiles, reviews, and discusses the available data and information on the electrical resistivity of aluminum and manganese and presents the recommended values resulting from critical evaluation, correlation, analysis, and synthesis of the available data and information. The recommended values presented are uncorrected and also corrected for the thermal expansion of the material and cover the temperature range from 1 K to above the melting point into the molten state for aluminum and to 700 K for manganese. The estimated uncertainties in most of the recommended values are about $\pm 2\%$ to $\pm 5\%$.

Key Words: aluminum; manganese; conductivity; critical evaluation; data analysis; data compilation; data synthesis; electrical conductivity; electrical resistivity; elements; metals; recommended values; resistivity.

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*Figures include the recommended values.

NOMENCLATURE

A	Constant in Eqs. (3b) and (8)
c	Impurity concentration
C	Constant in Eq. (3a)
e	Base of natural logarithm
h	Planck constant divided by 2π
k	Boltzmann constant
L	Length of specimen at T
L_0	Length of specimen at T_0
ΔL	$\Delta L = L - L_0$
M	Atomic weight
RRR	Residual resistivity ratio
T	Temperature
T_0	Reference temperature
x	$x = h\omega/kT$
a	Constant in Eqs. (7) and (8)
Δ	Deviation from the Matthiessen's rule
θ_D	Debye temperature
θ_R	Characteristic temperature for intrinsic electrical resistivity
ρ	Electrical resistivity
ρ_0	Residual electrical resistivity
ρ_e	Electrical resistivity due to electron-electron scattering
ρ_i	Intrinsic electrical resistivity
ω	Phonon angular frequency

1. INTRODUCTION

The principal objective of this project was to exhaustively compile, critically evaluate, analyze, and synthesize all the available data and information on the electrical resistivity of a large number of selected elements and to generate recommended values over a full range of temperature from 1 K to the melting point and beyond. The results on the electrical resistivity of aluminum and manganese are presented in this work (for manganese the recommended values cover the temperatures up to 700 K only), which is one in a series of similar works on the electrical resistivity of selected elements, some published¹⁻³. The comprehensive study of the electrical resistivity of the elements at the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) has been a continuation of a similar extensive work on the thermal conductivity of the elements⁴.

The general background information on this work is given in Section 2, which includes a brief introduction to the theory of the electrical resistivity of metals and a detailed explanation of the specifics and conventions used in the presentation of the data and information.

The experimental data and information and the recommended values for the electrical resistivity of the two elements are presented in Section 3. In the discussion of the electrical resistivity of each element, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties in the recommended values are stated. The recommended values uncorrected and corrected for the thermal expansion of the material are both presented in this section. The values cover the temperature range from 1 K to above the melting point for aluminum and to 700 K for manganese.

The last three sections are for acknowledgments, appendices, and references. There are two appendices given. The first appendix presents a logical organization of the methods for the measurement of electrical resistivity. The methods are designated with respective code letters and the same code letters are used in the 'Method Used' column of the Table of Measurement Information for indicating the experimental methods used by the various authors. The

second appendix presents conversion factors for the units of electrical resistivity, which may be used to convert easily the electrical resistivity values in the SI units given in this work to values in any of the several other units listed.

2. GENERAL BACKGROUND

2.1. Theoretical Background

It was found experimentally by Matthiessen^{5,6} that the increase in the electrical resistivity of a metal due to the presence of a small amount of another metal in solid solution is independent of the temperature. According to this Matthiessen's rule, the total electrical resistivity of an impure metal may therefore be separated into two additive contributions and written in the form

$$\rho(c,T) = \rho_0(c) + \rho_i(T) \quad (1)$$

where ρ_0 is the residual resistivity caused by the scattering of electrons by impurity atoms and lattice defects and is temperature-independent but dependent on the impurity concentration, c , and ρ_i is the temperature-dependent intrinsic resistivity arising from the scattering of electrons by lattice waves or phonons.

In reality, however, deviations from Matthiessen's rule do occur. Thus, in general the electrical resistivity of an impure metal is given by

$$\rho(c,T) = \rho_0(c) + \rho_i(T) + \Delta(c,T), \quad (2)$$

where Δ is the deviation from the Matthiessen's rule.

The intrinsic electrical resistivity which is due to scattering of electrons by phonons may be approximated by the Bloch-Grüneisen formula^{7,8}:

$$\rho_i = \frac{C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2} \quad (3a)$$

$$= A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2}, \quad (3b)$$

where C is a constant characteristic of the metal and proportional to the square of the electron-phonon interaction constant, M is the atomic weight, θ_R is a characteristic temperature of the metal which characterizes its intrinsic electrical resistivity in the same way as the Debye temperature, θ_D , characterizes its lattice specific heat, and $A \equiv C/M\theta_R$. The dimensionless variable of integration $x = h\nu/kT$, where h is the Planck constant divided by 2π , ν is the

phonon angular frequency, and k is the Boltzmann constant. The derivation of Eq. (3) is based on the simplifying assumptions that the Fermi surface is spherical, that the conduction electrons can be treated as free in the first approximation, that the spectrum of lattice vibrations is that of the Debye model, that the phonon distribution is essentially undisturbed by the scattering processes, and that electron-phonon Umklapp processes can be ignored. Consequently, it is perhaps most reasonable to expect the Bloch-Grüneisen formula to agree with experiment in the case of monovalent metals. Nevertheless, the intrinsic resistivity of many metals can be well represented by Eq. (3) over a wide temperature range by a suitable choice of θ_R and C , though no single values of θ_R can fit the data at all temperatures.

At low temperatures ($T \leq \theta_R/20$), Eq. (3a) reduces to

$$\rho_i = \frac{124.4C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5. \quad (4)$$

while at high temperatures ($T > \theta_R$), to a good approximation, it reduces to

$$\rho_i \approx \frac{C}{4M\theta_R} \left(\frac{T}{\theta_R} \right). \quad (5)$$

Thus it agrees with the experimental facts that at very low temperatures the intrinsic or ideal electrical resistivity (after subtracting ρ_0 from ρ) of most metallic elements is proportional to T^5 which is attributed to electron-phonon intraband scattering, and at high temperatures the resistivity of most metals increases approximately linearly with temperature.

In separating the electrical resistivity into its components, the temperature dependent part sometimes includes the electrical resistivity due to electron-electron scattering, ρ_e ; indeed, this is thought to be the dominant temperature-dependent term in transition metals at low temperatures. That is,

$$\rho = \rho_0 + \rho_e + \rho_i(T). \quad (6)$$

As in the case of the scattering of electrons by phonons, electron-electron collisions are of two types: normal processes in which the total wave vector is conserved, and Umklapp processes in which the total wave vectors before and after the collision differ by a reciprocal lattice vector. On the other hand, unlike electron-phonon Umklapp processes which are frozen out at

low temperatures if the Fermi surface is everywhere clear of the zone boundary, electron-electron Umklapp processes are not frozen out at low temperatures. Normal processes, involving the collision between two s-band conduction electrons, do not contribute directly to the electrical resistivity because they do not change the total momentum and thus have no effect on the current. Normal processes involving the scattering of an s-band conduction electron by a non-conducting d-band electron do contribute to the electrical resistivity, and are thought to be the dominant temperature-dependent resistive processes in transition elements and their alloys at very low temperatures, since their resistivities show the T^2 temperature dependence expected for electron-electron scattering rather than the T^5 temperature dependence expected for the intrinsic resistivity. This temperature dependence of the electrical resistivity due to electron-electron scattering:

$$\rho_e = aT^2 \quad (7)$$

comes about through the double application of the exclusion principle in the scattering processes; it applies to both the initial states and final states. In Eq. (7), a is a constant.

Umklapp processes between two conduction electrons do contribute to the electrical resistivity. Because these processes involve a reciprocal lattice vector, the wave functions of the electrons involved cannot be regarded as simple plane waves, but must be treated as true Bloch functions having the periodicity of the lattice. The results of this are to introduce into the expression for the resistivity the square of an interference factor. Apparently this factor is quite small, as the low temperature electrical resistivity of most ordinary metals does not show the T^2 temperature dependence expected for such a resistive mechanism.

Substituting Eqs. (7) and (3b) into Eq. (6) yields

$$\rho = \rho_0 + aT^2 + A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^x dx}{(e^x - 1)^2} . \quad (8)$$

Equation (8) has been used frequently in analyzing the experimental data and in generating the recommended values for the electrical resistivity at low temperatures.

2.2. Presentation of Data and Information

In each of the subsections in Section 3, electrical resistivity data and information for each element are presented in the following order:

- (1) A discussion text,
- (2) A table of recommended values,
- (3) A figure presenting experimental data as a function of temperature in a log-log scale (for manganese, due to the relatively small number of data sets, this figure is not given),
- (4) A figure presenting recommended values and selected experimental data (on which the recommendations were based) as a function of temperature in a log-log scale,
- (5) A figure presenting recommended values and selected experimental data (on which the recommendations were based) as a function of temperature in a linear scale,
- (6) A table giving measurement information on the experimental data presented in the figures, and
- (7) A table of experimental data for all the data sets listed in item 6 above.

In the discussion text on the electrical resistivity of each element, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties of the recommended values are stated.

The recommended values are for well-annealed high-purity and unoxidized specimens of the respective elements; however, those values for low temperatures are applicable only to the particular specimens having residual electrical resistivities as given at 1 K in the tables.

The recommended values uncorrected and corrected for the thermal expansion of the element are both given in the table. The uncorrected and corrected values are related by the following equation:

$$\rho_{\text{corrected}}(T) = \left[1 + \frac{\Delta L(T)}{L_0} \right] \rho_{\text{uncorrected}}(T). \quad (9)$$

where $\Delta L = L - L_0$ and L and L_0 are the lengths of the specimen at any temperature T and at a reference temperature T_0 , respectively. The thermal expansion correction for aluminum amounts roughly to about -0.5% to -0.9% at very low

temperatures, zero at room temperature, about 0.5% at 500 K, and about 1.6% near the melting point of the element. For manganese, the correction is about -0.3% at low temperature, zero at room temperature, and 0.8% at 500 K.

The recommended values in some cases are given with more significant figures than warranted, which is merely for tabular smoothness or for the convenience of internal comparison. Hence, the number of significant figures given in the table has no bearing on the degree of accuracy or uncertainty in the values; the uncertainty in the values is always explicitly stated.

In the figures, a data set consisting of a single data point is denoted by a number enclosed by a square, and a curve that connects a set of two or more data points is denoted by a ringed number. These data set numbers correspond to those listed in the accompanying tables providing measurement information and tabulating numerical data for each of the data sets. When several sets of data are too close together to be distinguishable, some of the data sets, though listed and tabulated in the tables, are omitted from the figure for the sake of clarity. The data set numbers of those data sets omitted from the figure are asterisked in both tables providing the measurement information and tabulating the experimental data.

The tables providing the measurement information contain for each set of experimental data the following information: data set number, reference number, author(s), year of publication, experimental method used for the measurement, temperature range covered by the data, name and specimen designation, specimen composition, specification and characterization, and information on measurement conditions, which are contained in the original paper. The experimental methods used for the measurement of the electrical resistivity are indicated in the column headed 'Method Used' in the table by the following code letters:

- A Direct-current potentiometer method
- B Direct-current bridge method
- C Alternating-current potentiometer method
- D AC bridge method
- K Direct heating method
- P Van der Paw method
- R Rotating magnetic field method

- This symbol means either that the method described by the author is not sufficient for assigning a specific code letter or that the use of a code letter would not convey enough of the information reported in the research document, and therefore the method used is described briefly in the last column of the table.

Details of these and other methods for the measurement of electrical resistivity may be found in the literature references given in Appendix 5.1, which presents a complete scheme for the classification and organization of the methods.

In the tables tabulating the experimental data, all the original data reported in different units have been converted to have the same units: the SI units $10^{-8} \Omega \text{ m}$. The recommended values generated are also given in the same units. Conversion factors for the units of electrical resistivity, which may be used to convert the electrical resistivity values in the SI units given in this work to values in other units, are given in Appendix 5.2.

3. ELECTRICAL RESISTIVITY DATA AND INFORMATION

3.1. Aluminum

There is a large body of data and information available on the electrical resistivity of aluminum. This includes data not only on very pure bulk material (indicated by a 5N purity, very large RRR of up to 46000, and very low residual resistivity, ρ_0 , of the order of $10^{-12} \Omega m$) but also on thin films as well as on effects such as those of cold-work, quenching, annealing, deformation, irradiation, and pressure. Over 190 data sets, mostly on the bulk material as a function of temperature, are presented in this work.

The information on specimen characterization and on the measurement condition for each of the data sets is given in Table 2. The data sets are tabulated in Table 3 and shown partially in Fig. 1. Only those data sets used in the recommendation procedure are shown in Figs. 2 and 3.

The work reported in the last several years (data sets 1-67) is concentrated on the study of the low-temperature behavior of the electrical resistivity and the origin of the so-called DMR (deviation from Matthiessen's rule). It has been reported that various scatterers such as impurities, dislocations, and surfaces (as in the size effect) can change the temperature-dependent resistivity substantially and can produce large DMR. Many of the data sets reported in Tables 2 and 3 can be rejected as the basis for estimation of the electrical resistivity of pure aluminum because of the impurity content, cold work, or inadequate annealing of the samples. Other data sets are for specimens subjected to procedure intended to produce oxidized surface layers. Most of the available data appears to be uncorrected for thermal expansion of the samples, although this correction amounts to 1.6% near the melting point. Among the data sets reported in Table 2, only the data of Cook et al.²² (data set 69), Radenac et al.⁴⁴ (data set 104), Wilkes⁵³ (data set 115) and of Simmons and Balluffi⁷⁴ (data set 150) are explicitly stated to have been corrected for thermal expansion, and the opposite has been assumed in all other cases.

Deviations from Matthiessen's rule are quite significant in aluminum and have been carefully studied. Ribot et al.⁹ (data sets 1-21) concluded that Matthiessen's rule is obeyed below 4.2 K. However, their studies do not extend above this temperature. Another complicating factor is the importance of

surface scattering for the resistance at low temperatures of pure samples in the form of foils or wires of diameter much less than 1 mm. This size-dependent contribution to the measured resistance, which is about proportional to T^2 , is comparable to the temperature-dependent resistance at 2 K. Its role in the reported low-temperature behavior of electrical resistivity for aluminum has been the subject of study and disagreement. It is attributed to a change in the electron distribution near the surface and is reported by van der Mass et al.⁹⁷ to depend only on the surface resistivity. Sample-dependent anomalies complicate the study of the temperature dependence of the size effect below 4 K.

There has been an active interest in measuring and analyzing the bulk resistivity of aluminum in the low-temperature range. Sambles et al.⁹⁸ have given an extensive list of effective single-power laws that have been used in representing this resistivity over various temperature ranges below 60 K. Generally speaking, the temperature dependent part of the resistivity drops from T^5 dependence above 20 K to a T^2 dependence around 2 K. The careful studies of Ribot et al.⁹ (data sets 1-21), based on their measurements of aluminum samples with RRR up to 40600, yield a temperature dependent resistivity that can be represented by $AT^2 + BT^5$ below 2.2 K, with the T^2 term dominant. This form has been found to be useful by others over a considerably wider temperature range. The T^2 -dependence around 2 K has been confirmed by Garland and Van Harlingen¹³ (data sets 48-54), van Kempen et al.⁹⁹, and Boysel et al.¹⁰⁰. According to the elementary theory of electron-electron scattering in metals, it would give rise to a T^2 term in the resistivity, and the observed T^2 -dependence of the electrical resistivity in aluminum around 2 K is commonly attributed to this scattering. The observed T^2 term is, however, much larger than that given by the simple theory of electron-electron scattering. A promising elaboration of the theory has been suggested by MacDonald¹⁰¹. Other researchers who deal with this subject are Nakamichi and Kino¹⁰ (data sets 22-28), Babic et al.¹⁸ (data sets 60,61), Aleksandrov and D'yakov⁶⁸ (data sets 139-141), Senoussi and Campbell³² (data sets 85,86), and Refs. 104-108.

The recommended values for the electrical resistivity at low temperatures are based on the data of Nakamichi and Kino¹⁰ (data sets 22-28) who studied samples of such high purity that surface scattering of the carriers became a significant factor in small wires or thin foils. Specifically, their values

for the resistivity of bulk aluminum (data set 28), derived by extrapolating their results for thicker and thicker samples, were used as the basis for the recommended values below 40 K. These are the representative values to be expected for bulk samples with ρ_0 of the order $10^{-12} \Omega \text{ m}$, or RRR approaching 27000. From 40 to 400 K, the recommended values follow closely the data of Cook et al.²² (data set 69), Seth and Woods⁴⁵ (data set 105), Wilkes⁵³ (data set 115) Moore et al.⁶⁰ (data set 125), and of Simmons and Balluffi⁷⁴ (data set 150). From 400 K to the melting point, the recommended values are based on the reasonably concordant (allowing for the different treatments of thermal expansion) results of Kedves et al.²⁸ (data set 79), Redenac et al.⁴⁴ (data set 104), and of Logunov and Zverev⁴⁸ (data set 109). It is worth noting that their data show a progressive increase in the electrical resistivity values above the expected linearly extrapolated values above 700 K. This was attributed by Simmons and Balluffi⁷⁴ to scattering by thermally generated point defects of the type which add atomic sites (vacancy-type defects).

There are about 15 data sets available for the electrical resistivity of aluminum in the liquid region. The temperature range covered by these data sets is from 933 to 1973 K. There is a general agreement ($\pm 5\%$) between most of the data sets. The recommended values in the liquid region are based on these data sets, giving weight to the data of Romanova and Persson³⁵ (data set 89), Levin et al.⁴⁰ (data set 95), Powell et al.⁶³ (data set 130), Roll et al.⁷⁸ (data set 157), and of Matuyama⁸⁸ (data set 181).

The recommended values for the electrical resistivity given in Table 1 and shown in Figs. 2 and 3 are for well-annealed unoxidized aluminum of purity 99.99% or higher, but those below 40 K apply specifically to samples with $\rho_0 = 1.00 \times 10^{-12} \Omega \text{ m}$. The table gives both values uncorrected and corrected for thermal expansion, while Figs. 2 and 3 show only the uncorrected recommended values along with the experimental data which were used to generate these values. Thermal expansion values needed to carry out thermal expansion correction were taken from ref. 109. The uncertainty in the recommended values is estimated to be within $\pm 1\%$ below 400 K, $\pm 2\%$ from 400 K up to the melting point, and about $\pm 3\%$ for the liquid.

As mentioned earlier, electrical resistivity measurements for aluminum reported in the literature are not only for bulk material but also for aluminum under various physical as well as thermal conditions. Additional information

is available in refs. 110-188. In the following paragraphs, an attempt is made to sort the source documents to highlight important effects.

The electrical resistivity studies at low temperature of thin films is of great interest to many researchers. The main purpose of the study appears to study so-called 'size effect.' Some of the works are cited above. The effect of grain boundary scattering on the electrical resistivity was reported by Bandyopadhyay and Pal¹⁸⁹ and by Andrews et al.¹⁹⁰. Additional information on the thin films in general is reported in refs. 191-211.

Properties such as specific heat as well as electrical resistivity show a progressive increase above the linearly extrapolated values at high temperatures. As mentioned earlier, this increase is ascribed to scattering by thermally generated point defects. Several semiempirical approaches to calculate contribution to vacancy-type defects have been proposed by various investigators. In addition to the study of Simmons and Balluffi⁷⁴ reported here, the readers are directed to refs. 212-230.

The lattice defects of solids induced at low temperature by irradiation have received considerable attention in the recent years. These studies are reported in refs. 231-250. The effect of deformation on the electrical resistivity is also an equally well investigated area. Interested readers may refer to refs. 251-269 for information on the electrical resistivity of deformed aluminum. Last but not least, magnetic field effects are reported in refs. 270-277, effects of heat treatment, quenching, and cold-working are given in refs. 278-290, and effects of high pressure are discussed in refs. 291-296.

TABLE 1. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF ALUMINUM^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{ m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.000100	0.000100	700	7.350	7.322
2	0.000102	0.000102	800	8.700	8.614
4	0.000109	0.000109	900	10.18	10.005
7	0.000139	0.000140	933.52	10.74(s)	10.565(s)
10	0.000193	0.000192	933.52		24.77(l)
15	0.000346	0.000345	1000		25.88
20	0.000755	0.000748	1100		27.46
25	0.00187	0.00186	1200		28.95
30	0.00453	0.00451	1300		30.38
40	0.0181	0.0180	1400		31.77
50	0.0478	0.0476	1500		33.11
60	0.0959	0.0955	1600		34.40
70	0.1624	0.1618	1700		35.69
80	0.245	0.2439	1800		36.93
90	0.339	0.338	1900		38.18
100	0.442	0.440	2000		39.34
150	1.006	1.003			
200	1.587	1.584			
250	2.157	2.155			
273	2.417	2.417			
293	2.650	2.650			
300	2.733	2.733			
400	3.866	3.875			
500	4.995	5.020			
600	6.130	6.122			

^aThe values are for well-annealed aluminum of purity 99.99% or higher, but those below 40 K apply specifically to samples with $\rho_0 = 1.00 \times 10^{-12} \Omega \text{ m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation.

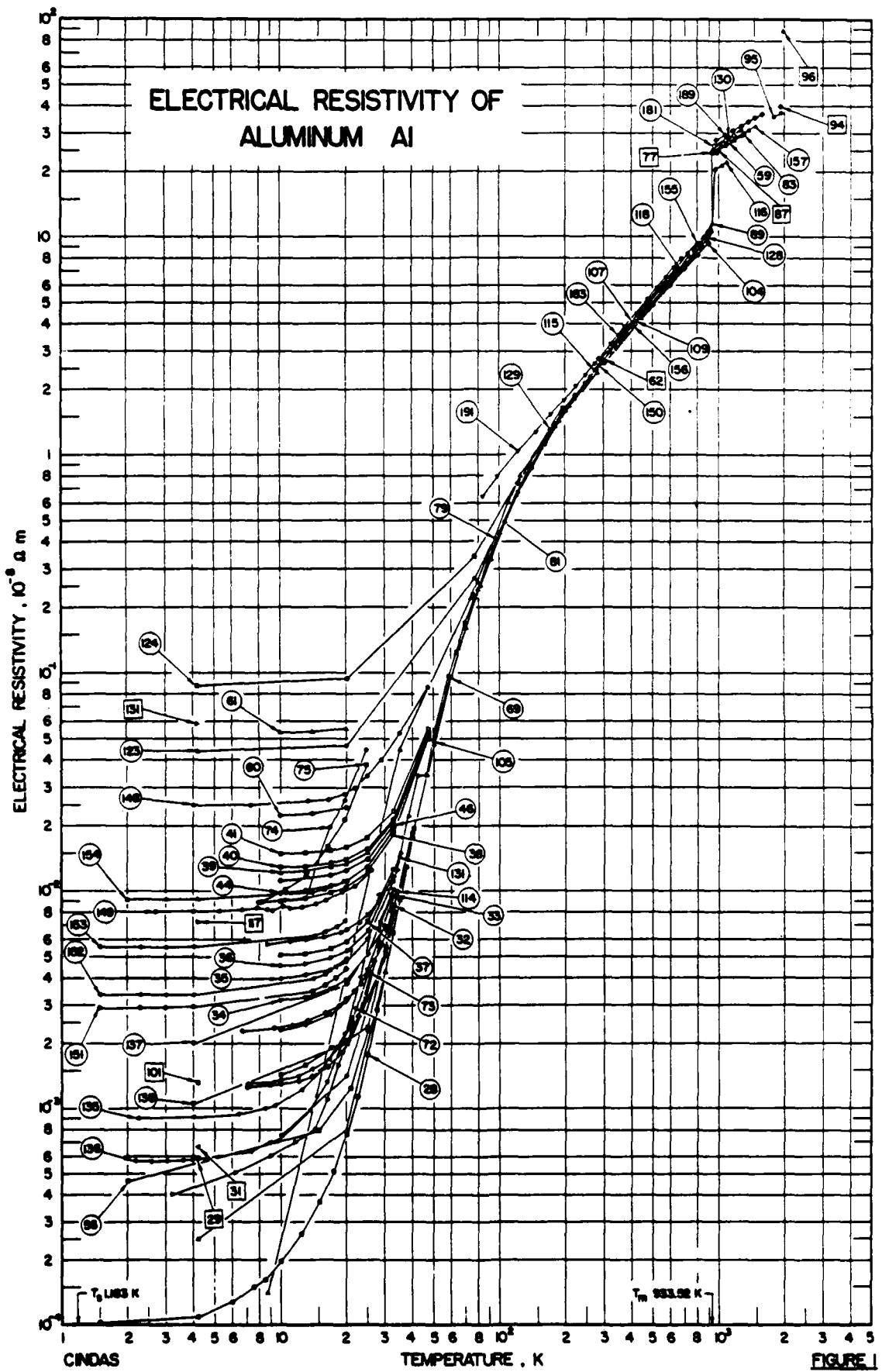
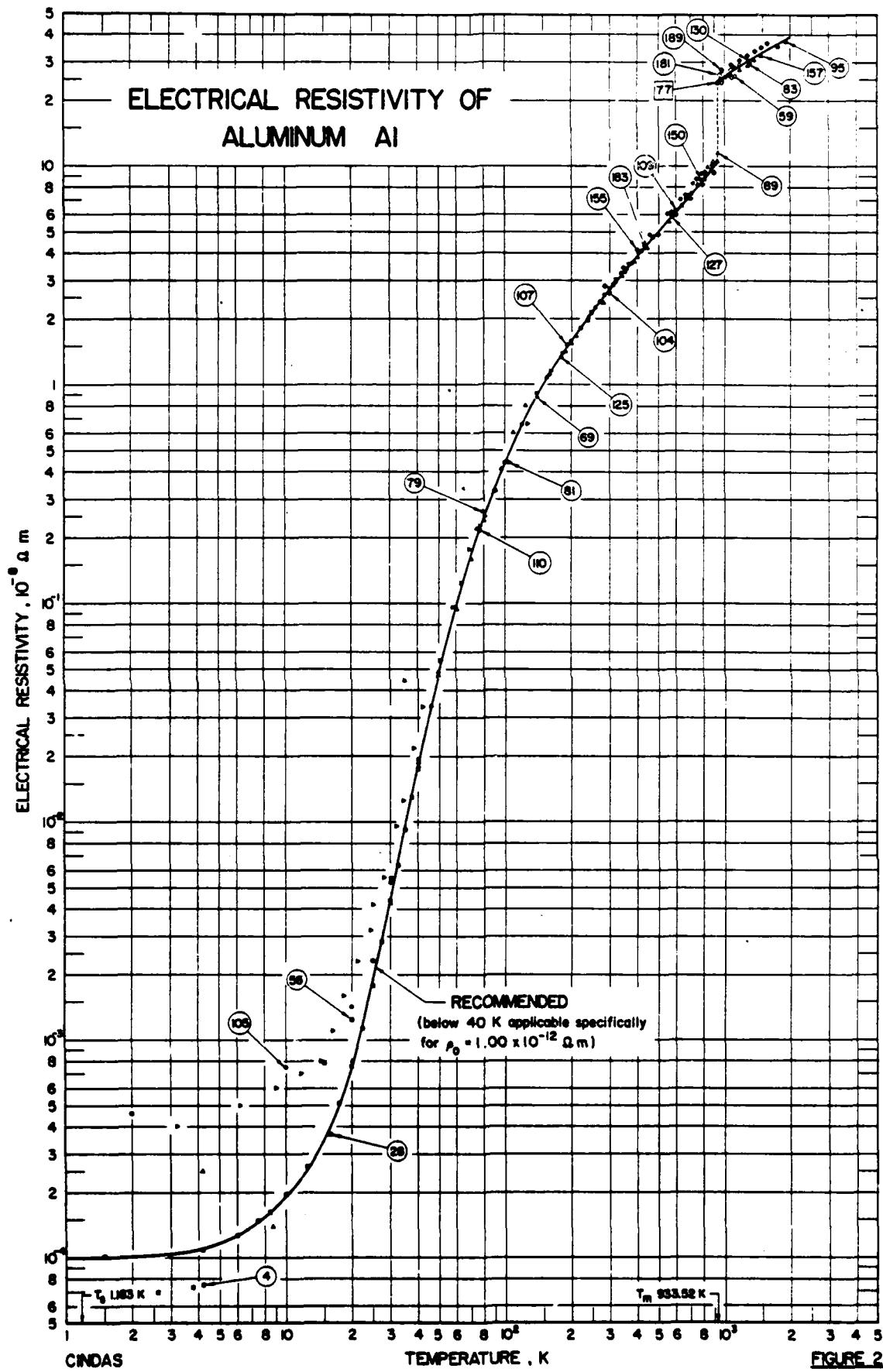


FIGURE 1

**FIGURE 2**

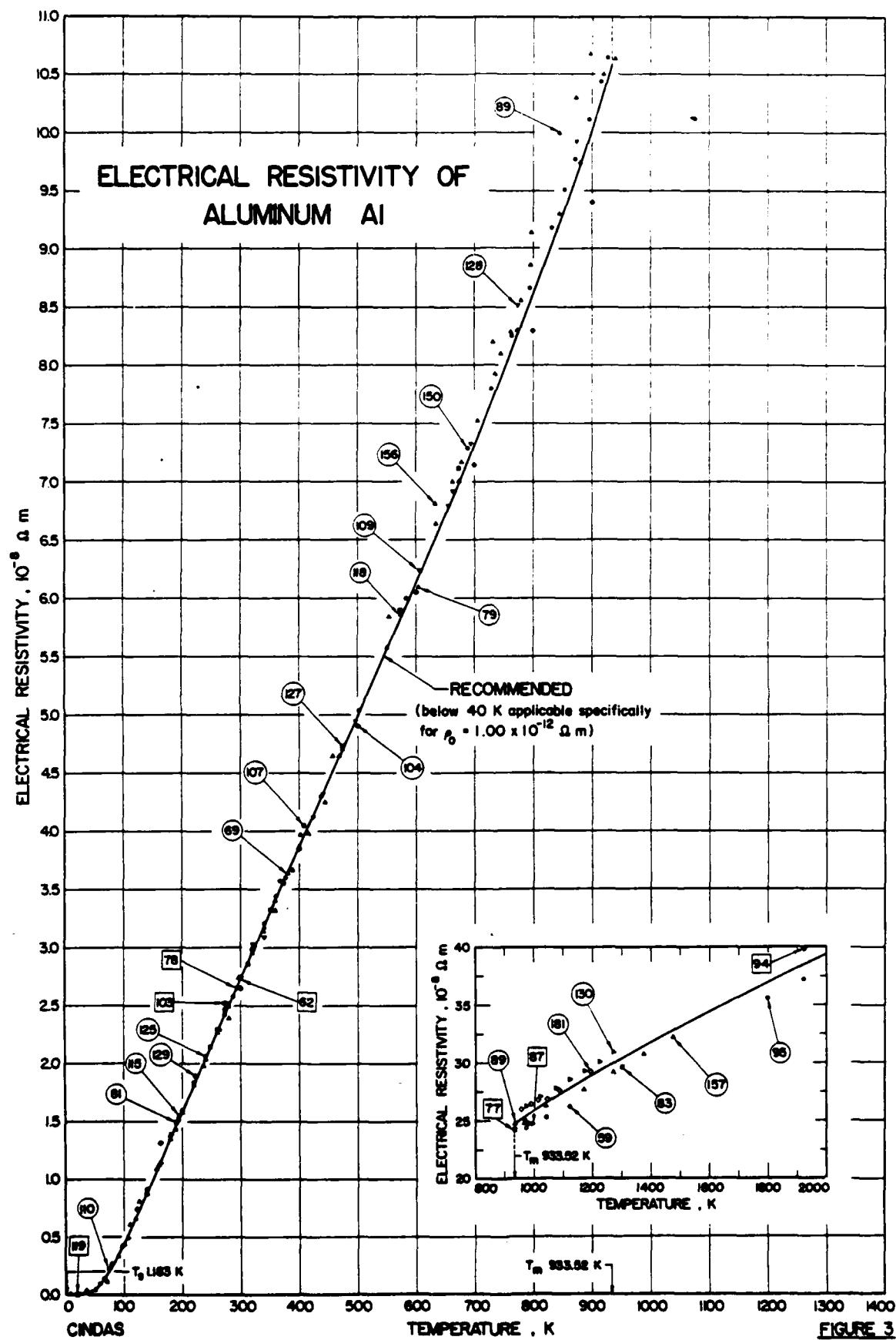


FIGURE 3

TABLE 2. MEASUREMENT INFORMATION OF THE ELECTRICAL RESISTIVITY OF ALUMINUM Al

Data Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
1* 9	Ribet, J.H.J.M., Basse, J., van Kempen, B., van Vucht, R.J.M., and Wyder, P.	1981	+	1.600-2.171	Sample 1	High purity specimen: nominal impurity <0.5 ppm; $\rho_0 = 0.0000928 \times 10^{-9} \Omega m$; RR = 29000; 1.4 mm diam. and about 1.5 m long cylindrical wire wound in double helix around quartz cylinder; before mounting, samples were cleaned in 40% NaOH solution to facilitate spot welding to 1 mm diam. ultrapure aluminum potential leads; welds were made with minimum electrical energy needed to achieve mechanical stability and showed no extra oxide formation; after annealing, test weld had resistance $< 5 \times 10^{-9} \Omega$ at 4.2 K; samples were annealed in dry hydrogen (510 ppm water) at 1 atm for 1 h at 773 K and 1 h at 673 K and cooled slowly to room temperature; lead wires were superconducting, attached using superconducting solder, $T_c = 1.16$ K; measurement utilizing superconducting flux gated galvanometer and current comparator with optimal precision of 0.1 ppm; series "a" data.
2* 9	Ribet, J.H.J.M., et al.	1981	+	1.298-3.842		Same as above except measurements designated as series "b".
3* 9	Ribet, J.H.J.M., et al.	1981	+	1.600-2.171		Same as above except measurements designated as series "c".
4 9	Ribet, J.H.J.M., et al.	1981	+	2.631-4.221	Sample 2	Same as in data set 1 except sample diam. 3.0 mm; $\rho_0 = 0.0000667 \times 10^{-9} \Omega m$; RR = 40,600; measurements designated as series "a".
5* 9	Ribet, J.H.J.M., et al.	1981	+	2.362-3.997		Same as above except measurements designated as series "b".
6* 9	Ribet, J.H.J.M., et al.	1981	+	4.134-4.224		Same as above except measurements designated as series "c".
7* 9	Ribet, J.H.J.M., et al.	1981	+	1.180-2.172		Same as above except measurements designated as series "d".
8* 9	Ribet, J.H.J.M., et al.	1981	+	2.578-4.220	Sample 3	Same as in data set 1 except sample diam. 3.0 mm; $\rho_0 = 0.0013 \times 10^{-9} \Omega m$; RR = 21000; nominal impurity <5 ppm; measurements designated as series "a".
9* 9	Ribet, J.H.J.M., et al.	1981	+	1.950-2.80		Same as above except measurements designated as series "b".
10* 9	Ribet, J.H.J.M., et al.	1981	+	1.292-1.900		Same as above except measurements designated as series "c".
11* 9	Ribet, J.H.J.M., et al.	1981	+	1.253-1.451		Same as above except measurements designated as series "d".
12* 9	Ribet, J.H.J.M., et al.	1981	+	2.049-2.100	Sample 4	Same as in data set 1 except nominal impurity <8 ppm; sample diam. 3.0 mm; $\rho_0 = 0.000292 \times 10^{-9} \Omega m$; RR = 9300; measurements designated as series "d".

Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
13 ^a 9	Ribot, J.H.J.M., et al.	1981	+	3.183-4.133		Same as above except measurements designated as series "b".
14 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.501-4.221		Same as above except measurements designated as series "c".
15 ^a 9	Ribot, J.H.J.M., et al.	1981	+	4.209		Same as above except measurements designated as series "d".
16 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.254-1.601		Same as above except measurements designated as series "e".
17 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.522-4.218	Sample 5	Nominal impurity <100 ppm; $\rho_0 = 0.01068 \times 10^{-8} \Omega \cdot m$; RRR = 255; cylindrical wire 2.0 mm diam. and 10 cm long; cleaned in NaOH solution, annealed in hydrogen as described in data set 1 and then recleaned in solution; ultrapure, 3 cm long aluminum potential leads were then spot-welded to sample 2 cm in from each end; mounting of sample was achieved as described in data set 1.
18 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.284-4.200	Sample 6	Same as above (data set 17) except impurity unknown; $\rho_0 = 0.01106 \times 10^{-8} \Omega \cdot m$; RRR = 245; specimen diam. 1.0 mm.
19 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.224-4.206	Sample 7	Intermediate purity sample, impurity <10 ppm; $\rho_0 = 0.000663 \times 10^{-8} \Omega \cdot m$; RRR = 4100; samples were spark-cut from aluminum sheet 1 mm thick, 10 cm long, and 1 mm wide containing four tabs 1 mm wide and 2 mm long located approximately symmetrically on the sample about 1 cm in from each end; cleaned in NaOH solution; annealed in air; potential contacts were soldered to ends of two tabs on the same side of the sample.
20 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.371-4.229	Sample 8	Same as above (data set 19) except $\rho_0 = 0.000601 \times 10^{-8} \Omega \cdot m$; RRR = 4500; sample annealed in hydrogen for 22 h.
21 ^a 9	Ribot, J.H.J.M., et al.	1981	+	1.241-4.211	Sample 9	Same as above (data set 19) except $\rho_0 = 0.002245 \times 10^{-8} \Omega \cdot m$; RRR = 1100; sample left unannealed.
22 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-42		Specimen made from block (10 x 20 x 90 mm ³) cut from zone refined polycrystalline Al bar (RRR = 26000); thickness 0.019 mm x 5 mm (reduced thickness 0.019 mm based on 2 x cross section divided by perimeter); specimen annealed for 3 h at 600°C in air and then cooled down in furnace; RRR = 1692; data taken from figure.
23 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-43		Similar to the above except thickness 1.494 mm and width 2.96 mm (reduced thickness 0.986 mm); RRR = 1730; values are fairly close to the bulk values calculated from data for strips using Fuchs-Sondheimer relation; data taken from figure.
24 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-35		Similar to the above except thickness 0.1955 mm and width 3.17 mm (reduced thickness 0.184 mm); RRR = 7523; data taken from figure.

^aNot shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM A1 (continued)

Data Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
25 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-41		Similar to the above except thickness 0.101 mm and width 4.66 mm (reduced thickness 0.099 mm); RR = 4697; data taken from figure.
26 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-42		Similar to the above except thickness 0.039 mm and width 5 mm (reduced thickness 0.039 mm); RR = 2717; data taken from figure.
27 ^a 10	Nakamichi, I. and Kino, T.	1980	A	1-42		Similar to the above except thickness 0.030 mm and width 5 mm (reduced thickness 0.030 mm); RR = 2041; data taken from figure.
28 10	Nakamichi, I. and Kino, T.	1980	A	1-40		Values for bulk material based on their measurements for 0.0195-1.484 mm thick strips of zone refined aluminum bar of bulk RR = 26600 and Yoch-Sondheimer relation; the values are fairly close to the values for 1.484 mm thick strip.
29 11	Kim, S.H. and Wang, S.T.	1978	A	4.2	Aluminum #1	99.99% Al; polycrystalline supplied by D. Koop of Alcoa; 0.7 cm diam. x 3.5 cm long; soft shouldered on both ends with copper bars 1.8 cm diam. x 7.5 cm long; resistivity obtained following relationship: $\rho(\epsilon, B) = \rho_0 + \rho_4 (\epsilon) + \rho_m (B)$ (ϵ & B have no significance since ϵ was considered at zero strain (ϵ) and zero magnetic field (B)); data taken from figure; reported error 10%.
30 ^a 11	Kim, S.H. and Wang, S.T.	1978	A	4.2	Aluminum #3	Similar to above specimen.
31 11	Kim, S.H. and Wang, S.T.	1978	A	4.2	Aluminum #4	Similar to above specimen.
32 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	99.999% Al; obtained from Koch-Light (type 8013 h, batch 1); 0.508 mm diam.; reduced by rolling and drawing through diamond dies to various diameters, and through a varying number of dies which accounts for reducing specimen diam. by 112 and changes in ρ_0 ; number of dies zero for this specimen; annealed at 340°C for 3 h; $\rho_0 = 0.001306 \times 10^{-6} \Omega \cdot m$; values calculated from graphically extracted values for ρ_T , temperature dependent resistivity.
33 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	Same as above except $\rho_0 = 0.00222 \times 10^{-6} \Omega \cdot m$; number of dies is 1.
34 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	Same as above except $\rho_0 = 0.00309 \times 10^{-6} \Omega \cdot m$; number of dies are 2.
35 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	Same as above except $\rho_0 = 0.00391 \times 10^{-6} \Omega \cdot m$; number of dies are 3.
36 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	Same as above except $\rho_0 = 0.00447 \times 10^{-6} \Omega \cdot m$; number of dies are 4.
37 12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(1)	Same as above except $\rho_0 = 0.00499 \times 10^{-6} \Omega \cdot m$; number of dies are 5.

^aNot shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. No.	Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
38	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-48	Al(2)	Similar to above specimen; $\rho_0 = 0.00874 \times 10^{-6} \Omega\text{m}$; number of dies are zero; run I.
39	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-48	Al(2)	Same as above except $\rho_0 = 0.0121 \times 10^{-6} \Omega\text{m}$; number of dies is 1.
40	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-48	Al(2)	Same as above except $\rho_0 = 0.0127 \times 10^{-6} \Omega\text{m}$; number of dies are 2.
41	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-48	Al(2)	Same as above except $\rho_0 = 0.0147 \times 10^{-6} \Omega\text{m}$; number of dies are 4.
42*	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-48	Al(2)	Same as above except $\rho_0 = 0.0148 \times 10^{-6} \Omega\text{m}$; number of dies are 6.
43*	12	Rowlands, J.A. and Woods, S.B.	1978	B	13,20	Al(2)	Similar to above specimen; $\rho_0 = 0.00877 \times 10^{-6} \Omega\text{m}$; number of dies are zero; run II.
44	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(2)	Same as above except diam. = 0.494 mm; $\rho_0 = 0.00963 \times 10^{-6} \Omega\text{m}$; number of dies are 1.4.
45*	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(2)	Same as above except diam. = 0.482 mm; $\rho_0 = 0.0102 \times 10^{-6} \Omega\text{m}$; number of dies are 1.2.
46	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(2)	Same as above except diam. = 0.469 mm; $\rho_0 = 0.0111 \times 10^{-6} \Omega\text{m}$; number of dies are 1.4.
47*	12	Rowlands, J.A. and Woods, S.B.	1978	B	10-33	Al(2)	Same as above except diam. = 0.458 mm; $\rho_0 = 0.01174 \times 10^{-6} \Omega\text{m}$; number of dies is 1.
48*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.35-6.46	Al-1	Pure, polycrystalline 30 mm diam. rods; $\rho_0 = 0.0002057 \times 10^{-6} \Omega\text{m}$; several normal resistance ratio = 12000; annealed in air at 550°C for several hours; RRR = 1232; values calculated from graphically reported $\rho_T - \rho_0$ vs T values; voltage measured using SQUID detector.
49*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.5-4.0	Al-2	Similar to above specimen; values are calculated from $\rho = \rho_0 + A T^2$ using $\rho_0 = 0.0002361 \times 10^{-6} \Omega\text{m}$, and $A = 5.4 \pm 0.4 \times 10^{-4} \text{n} \Omega \text{cm K}^{-2}$.
50*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.5-4.0	Al-3	Similar to above specimen; $\rho_0 = 0.0002012 \times 10^{-6} \Omega\text{m}$; RRR = 12588; $A = 5.7 \pm 0.4 \times 10^{-4} \text{n} \Omega \text{cm K}^{-2}$.
51*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.5-4.0	Al-4	Similar to above specimen but cold-worked after annealing; $\rho_0 = 0.0006195 \times 10^{-6} \Omega\text{m}$; RRR = 4201, $A = 6.7 \times 10^{-4} \text{n} \Omega \text{cm K}^{-2}$.
52*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.5-4.0	Al-5	Similar to above specimen; $\rho_0 = 0.0000519 \times 10^{-6} \Omega\text{m}$; RRR = 5030, $A = 5.2 \times 10^{-4} \text{n} \Omega \text{cm K}^{-2}$.
53*	13	Carlson, J.C. and Harlingen, D.J.	1978	A	1.5-4.0	Al-6*	Similar to Al-1; $\rho_0 = 0.0000944 \times 10^{-6} \Omega\text{m}$; RRR = 25999, $A = 4.3 \times 10^{-4} \text{n} \Omega \text{cm K}^{-2}$.

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
54 ^a 13	Carland, J.C. and Hartligen, D.J.	1976	A	1.5-4.0	Al-6b	Similar to above specimen but cold-worked after annealing; $\rho_0 = 0.00004638 \times 10^{-8} \Omega\text{m}$; RR = 5625, $A = 4.6 \times 10^{-4} \text{ m}^2 \text{ cm}^{-3} \text{ K}$. No details given.
55 ^a 14	Masovic, D.L. and Zalovic, S.	1976	A	933		99.999 Al; RR = 5900; $\rho_0 = 0.000046 \times 10^{-8} \Omega\text{m}$; values calculated from graphically reported $\rho_T - \rho_0$ values which are temperature dependent resistivity.
56 15	Kloptin, N.N., Panova, G.N., and Semolov, B.N.	1977	A	2-295		99.999 Al; zone refined specimen wires 0.6 mm in diam.; annealed in vacuum at 600°C for 2 days; all specimens chemically etched and rinsed with distilled water; $\rho_0 = 0.0000448 \times 10^{-8} \Omega\text{m}$; measurement done with SQUID galvanometer with voltage sensitivity $\pm 10^{-15}$ V; heating effects negligible; data extracted from figure; a main source of error was the specimen size; SQUID detector used; uncertainty about 1%.
57 ^a 16	Pujita, T. and Otsuka, T.	1977	A	1.51-9.72	Al-4	99.999 Al; zone refined specimen wires 0.6 mm in diam.; annealed in vacuum at 600°C for 2 days; all specimens chemically etched and rinsed with distilled water; $\rho_0 = 0.0000448 \times 10^{-8} \Omega\text{m}$; measurement done with SQUID galvanometer with voltage sensitivity $\pm 10^{-15}$ V; heating effects negligible; data extracted from figure; a main source of error was the specimen size; SQUID detector used; uncertainty about 1%.
58 ^a 16	Pujita, T. and Otsuka, T.	1977	A	1.50-9.09	Al-1a	Similar to the above specimen except it was cold-worked; sandwiched between clean Al sheets and rolled to 0.3 mm thick plate form; $\rho_0 = 0.001355 \times 10^{-8} \Omega\text{m}$. No details given; liquid state specimen; data extracted from figure.
59 17	Kita, M., Steinemann, S., Knauf, H.U., and Giechrodt, H.J.	1977	C	933-1122		No details given; liquid state specimen; data extracted from figure.
60 18	Babic, E., Kresek, R., and Ocho, M.	1976	A	10-20		99.999 Al from Koch Light; temperature controlled by helium exchange gas and by resistance heater; $\rho_0 = 0.022 \times 10^{-8} \Omega\text{m}$.
61 18	Babic, E., et al.	1976	A	10-20		Similar to above except $\rho_0 = 0.053 \times 10^{-8} \Omega\text{m}$.
62 19	Kawata, S.	1976	A	300	VIII-1	99.999 Al; zone refined; $\rho_0 = 0.000193 \times 10^{-8} \Omega\text{m}$.
63 ^a 20	Krevet, B. and Schaefer, W.	1976	A	4.2-32	Sample I	Pure; polycrystalline; from Vereinigte Aluminiumwerke, AG, Bonn; Al tape samples 0.3 x 6 mm ² cross-section; liquid hydrogen cryostat used; RR = 2200; data extracted from figure.
64 ^a 20	Krevet, B. and Schaefer, W.	1976	A	4.2-32	Sample II	Similar to above specimen; RR = 3800.
65 ^a 20	Krevet, B. and Schaefer, W.	1976	A	4.2-32	Sample III	Similar to above specimen; RR = 5600.
66 ^a 20	Krevet, B. and Schaefer, W.	1976	A	5.5-32	Sample IV	Similar to above specimen; RR = 8900.
67 ^a 20	Krevet, B. and Schaefer, W.	1976	A	4.2-32	Sample VI	Similar to above specimen; RR = 13900.
68 ^a 20	Hartwig, K.T. and Worsala, F.J.	1976	A	273		Pure; melted in induction furnace in high purity graphite crucibles under argon; ingots from the melt (1 in. diam.) were extruded to 1/4 in. diam.; specimens were then cut to 2 in. lengths and homogenized in air at 813 K for 12 h, then water quenched and immersed in liquid nitrogen for storage.

^a Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
69 22	Cook, J.G., Moore, J.P., Matsumura, and Van der Mar, M.P.	1975	A	4.2-400		99.999 Al; specimens purchased from Comilco Ltd., Guelph, Ontario; three samples measured with three techniques; sample with RRR = 11,000 annealed at Comilco Ltd.; sample with RRR = 8500 annealed at NBC; sample with RRR = 950 of commercial purity; data extracted from tabulated values which were obtained by passing a smooth curve approximately midway between the high and low results for the pure specimens; data reported were corrected for thermal expansion; author's estimated uncertainty 0.8%.
70 ^a 23	Rapp, O. and Fogelholm, A.	1975	A	318	Sample 1	Pure; <4 ppm of transition metal impurities and <36 ppm total impurities; rolled and drawn into wire 0.25 mm diam.; annealed at 450°C for 6 h.
71 ^a 23	Rapp, O. and Fogelholm, A.	1975	A	318	Sample 2	Similar to above specimen.
72 24	Rowlands, J.A. and Woods, S.B.	1975	B	7-26	Al 1 Type 8013h	99.999 Al from Koch-Light; 1 mm diam. wires reduced in diam. in stages by drawing through dies to final diam. of 0.02 in.; annealed at 340°C for 3 h in vacuum to remove physical defects and inhibit growth of very large crystallites which would prevent uniform drawing; $\rho_0 = 0.00124 \times 10^{-8} \Omega\text{m}$; values obtained from graphically reported temperature dependent electrical resistivity, ρ_T .
73 24	Rowlands, J.A. and Woods, S.B.	1975	B	7-26	Al 1	Same as above except plastically elongated at room temperature by amounts 5-300% by drawing them through dies, or, for small strains, stretching them.
74 24	Rowlands, J.A. and Woods, S.B.	1975	B	7-25	Al 2	Similar to the above annealed specimen except $\rho_0 = 0.0098 \times 10^{-8} \Omega\text{m}$; data extracted from figure.
75 24	Rowlands, J.A. and Woods, S.B.	1975	B	8-25	Al 2	Same as above except cold-worked to the smallest value of ρ_T ; data extracted from figure.
76 ^a 25	Konetsu, S. and Kino, T.	1975	A	4.2-300		99.999 Al supplied by Sumitomo Chemical Co., Ltd.; zone-refined; polycrystalline wire of 1 mm diam.; RRR = 12200-16200.
77 26	Srivastava, S.K.	1975		938		No details given.
78 27	Bradley, J.M. and Stringer, J.	1974	A	293		99.999 Al; cold rolled to a thickness of 0.5 mm from which rectangular specimen (5 mm x 40 mm) was cut; specimen was solution treated at 500°C and water quenched immediately prior to measurement of resistivity.
79 28	Kodera, F.J., Gergely, L., and Hordeos, M.	1973	A	26.4-947.9		99.999 Al; 50 mm long (at low temp.), 100-1200 mm long (at high temp.); wound to form a coil on a mica sheet; cold drawn (0.8-1.0 mm diam.); annealed and homogenized at 620-630°C for 1 h; double chamber cryostat used; data extracted from figure; reported error ±1%.
80 ^a 29	Oemura, K., Hirotsu, Y., and Marukami, Y.	1973	A	4.2-77	5W Grade	99.999 Al; 59 grade; supplied by Asahi Metal Co.; RRR = 9700.

^aNot shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
81 29	Ozumara, K., Hirooka, Y., and Marubayashi, T.	1973	A	3.2-200	5N Grade	99.999 Al, 59 grade; wire specimen 0.5 mm in diam.; supplied by Asahi Metal Co.; RR = 9700; low temperature unpublished data from Nakamura, Furukawa, and Takamura; data extracted from figure.
82* 29	Ozumara, K., et al.	1973	A	4.2-77		0.175 Zn; specimen 50 mm x 4 mm with long projecting Hall probes and 70 µm thick; supplied by Sumitomo Mining Co.; cold-rolled, solution treated for 1 h at 450°C, cooled and held for 1 h at 300°C; quenched in water at 0°C, and immediately immersed in liquid nitrogen.
83 30	Stallard, J.M. and Davis, C.H., Jr.	1973			976,1302 NMC VP Grade	99.995 Al; rod 5.08 cm.
84* 31	Thompson, G.B. and Noble, B.	1973	A	74.98-266.5		High purity; cast under argon in an induction furnace; ingots were extruded, homogenized, and cold-rolled to 1.3 mm strip; data extracted from figure.
85* 32	Sauvageot, S. and Campbell, I.A.	1973	A	1.32-4.21	Commercial 5N Al	Commercial 5N Al wire (RR = 1200); $\rho_0 = 0.002409 \times 10^{-9} \Omega \text{m}$; geometrical factor of the order of 10^3 ; data taken from figure of $\rho - \rho_0 / \rho$ vs T .
86* 32	Sauvageot, S. and Campbell, I.A.	1973	A	2.98-4.19	Commercial 3N Al	Commercial 3N Al wire (RR = 65); $\rho_0 = 43.31 \text{ m}\Omega \text{cm}$; geometrical factor of the order of 10^4 ; data taken from figure of $\rho - \rho_0 / \rho$ vs T .
87 33	Korochkin, I.M. and Kazantsev, V.P.	1973		993		Pure; no other details are given.
88* 34	Endicott, J.E. and Rose, R.A.	1973		1120		Pure.
89 35	Bogdanov, O.V. and Perel'man, Z.V.	1973		842.5-1041.3		Pure aluminum specimen.
90* 36	Strelts, N.N., Gotsikhev, V.I., and Dronov, A.A.	1972		4.2,273		Single crystal; 60 x 4 x 3 mm; specimen axis along <110> direction; $\rho(273)$ calculated from resistance ratio of order of 6000 (assumed equal to resistivity ratio) and $\rho(4.2 \text{ K})$.
91* 37	Horak, J.A. and Blewitt, T.H.	1972	A	4.5-295		Polycrystalline wire specimen; approximately 5 cm long with a diam. of 0.025 cm.
92* 38	Callaretti, R.C. and Alfonsi, M.	1972	+	77		Bar of very common structural aluminum; 12 cm long, 9.5 mm diam; inductive method.
93* 38	Callaretti, R.C. and Alfonsi, M.	1972	+	77		Similar to the above; resistive method.
94 39	Levin, E.S., Ayushina, G.D., and Gal'd, P.V.	1972	R	1923	AV-000	99.99 Al; data taken from figure; contactless method.
95 40	Levin, E.S., et al.	1972	R	1923,1798	AV-00	99.99 Al; data taken from figure; reported error 7%; contactless method.

^aNot shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
96 41	Lewis, E.S. and Ayashina, G.D.	1972	R	1973	AV 000	99.99 Al; data taken from figure; contactless method.
97* 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 1	99.9999 Al; <0.03 at. ppm Ag, 0.1 at. ppm Cu, 0.5 at. ppm Fe, 0.1 at. ppm Mg, 0.5 at. ppm Si; from Comitaco American Inc.; ribbon shaped, 18 cm long, 0.080 cm wide, and 0.017 cm thick; annealed in air at 600 ± 5°C for zero h.
98* 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 2	Same as above except annealed for 5 h.
99* 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 3	Same as above except annealed for 20 h.
100* 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 4	Same as above except annealed for 23 h.
101 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 5	Same as above except annealed for 36 h.
102* 42	D'Malifi, R.J. and Siegel, R.Y.	1971	A	4.2	Specimen No. 6	Same as above except annealed for 48 h.
103 43	Alp, T., Brough, I., Sanderson, S.J., and Mattock, K.H.	1970	A	273		99.9999 Al; zone refined; 8 ppm impurities by weight; 0.508 mm diam. wire; quenched in ice water at 0°C from 200°C.
104 44	Radencic, A., Lacoste, H., and Bourg, C.	1970	R	300-900		99.995 Al; 0.0060 Mg, 0.0005 Fe, 0.0002 Cu, and 0.0002 Si; 4 mm diam. × 3 mm; expansion corrected; uncertainty ±3%; contactless method.
105 45	Seth, R.S. and Woods, S.B.	1970	A	10-295	Grade 5M	99.999 Al; polycrystalline; obtained from Consolidated Mining and Smelting Co. of Canada; 6 mm diam. rod drawn through steel dies to 1.5 mm diam., then etched, then drawn through diamond dies to 0.5 mm diam.; annealed for 12 h at 400°C in 10 ⁻² Torr atmosphere; electrical resistance ratio R(293 K)/R(4 K) = 4000; resistivity deduced from $\rho = \rho_0 + \rho_0 (\ln T / \ln 273.2 K) - 2.429 \mu\Omega \text{ cm}$, $\rho_0(273.2 K) = 0.0007 \mu\Omega \text{ cm}$, $\rho_0(273.2 K)$ extracted from table.
106* 45	Seth, R.S. and Woods, S.B.	1970	A	273.2		0.12 Mg; 6 mm diam. rods made by melting freshly cleaned pellets in evacuated sealed quartz tubes, then drawn through steel dies to 1.5 mm diam.; etched and drawn through diamond dies to 0.5 mm diam.; annealed at 400°C for 12 h in 10 Torr H ₂ atmosphere in close-fitting Pyrex container; residual resistivity 0.0487 μΩ cm.
107 46	Söhn, R. and Wachtel, E.	1969		194-408		99.997% Al; impurities 0.001 Cu, 0.001 Fe, 0.001 Si; cylindrical specimen 10 mm in diam.
108* 47	Subbarao, V.R. and Grossman, M.I.	1969		293		7 × 7 × 28 mm; measuring temperature assumed 20°C.

Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
109	48	Loganov, A.V. and Izverev, A.P.	1966	A	321-693	99.946 Al; 4 mm diam. x 100 mm; data taken from figure; not corrected for expansion of sample; reported error <0.5%.	
110	49	Wilkes, K.L. and Powell, R.W.	1966	A	77-273	99.9998 Al; polycrystalline; 0.5 ppm Cu, 0.5 ppm Si, 0.1 ppm Mg; obtained from Advanced Research Materials; 1.225 cm diam. x 10.16 cm.	
111*	50	Von Basswits, A. and Mitchell, R.M.	1966	A	4.6-90.3	99.999 Al.	
112*	51	Sharm, J.K.B.	1967	D	1.5-293	99.999 Al; polycrystalline wire specimen obtained from Aluminum Laboratories; RR = 664; 1 mm diam. x 70 cm long.	
113*	52	Stevenson, R.	1967		9-35	99.999 Al; wire obtained from Consolidated Mining and Smelting Co.; received extensive deformation in the wire-drawing process and further deformation when wound on mandrels of 0.5 in. diam. in making the samples for the experiment; mounted samples annealed at 150°C for 4 h; resistivity ratio = 476; residual electrical resistivity = 5.74 x 10^-11 Ω m; data extracted from smooth curve.	
114	52	Stevenson, R.	1967		9-35	Similar to the above except resistivity ratio = 1173; ρ₀ = 2.27 x 10^-11 Ω m; data extracted from smooth curve.	
115	53	Wilkes, K.L.	1967	A	78-298	99.9998 Al, 0.00005 Cu, 0.00005 Si, and 0.00001 Mg; 1.226 cm diam. x 10.16 cm long; obtained from Artesco Products Inc.; density 2.700 g/cm³; at 23°C, results corrected for thermal expansion by multiplying the room temperature dimensions by $(1 + \alpha_0 T)$ where α_0 is average coefficient of linear thermal expansion and T is the change from room temperature.	
116	54	Bosch, G. and Cimberroti, H.J.	1967	C	863-1080	No details given.	
117	55	Boatto, G., Bupo, M., and Risqueto, C.	1966	A	4.2	99.995 Al; the specimen was annealed in air for one day at 610°C, then quenched in iced salt water for less than a second; the measurement was taken using a Keithley nanovoltmeter, whose calibration was better than 3%.	
118	56	Mobili, D. and DeBucci, M.A.	1966	A	298-773	99.99 Al, <0.005 S, 0.003 Fe, <0.001 Mg, and <0.001 Zn; cylindrical specimen; annealed at 550°C for 2 h; reported error <1%.	
119	57	Mealy, H.H. and Soeis, A.	1966		20.4	99.9999 Al; specimen supplied by United Minerals Corp.; wire drawn to diam. of 0.0033 cm.	
120*	57	Mealy, H.H. and Soeis, A.	1966		20.4	99.995 Al; wire supplied by Aluminum Corporation of America; was drawn to 0.0033 cm diam.	
121*	58	Pawlak, F. and Rogalla, D.	1966	B	4-273	99.999 Al, 0.00024 Fe, 0.00019 Cu, 0.00015 Si, and 0.0003 remaining impurities; 2 mm diam. wire received, with work analysis, from Aluminum-Pütte Rheinfelden GmbH, Rheinfelden; electrical resistivity ratio p(273 K)/p(4.2 K) = 2210, p(293 K)/p(20.4 K) = 1130.	

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks
122*	58	Pavlek, F., and Rogalla, D.	1966	B	4-273	Very pure Al	99.994 Al, 0.0024 Cu, 0.0002 Si, and 0.0012 Fe; 2 mm diam. wire supplied by Vereinigte Aluminiumwerke AG, Bonn; annealed 1 h in argon at 300°C (authors report annealing temperature as 300°C in Fig. 5, but 400°C on P. 17 of their paper); cooling rate <50°C/h; electrical resistivity ratio $\rho(273\text{ K})/\rho(4.2\text{ K}) = 400$, $\rho(293\text{ K})/\rho(20.4\text{ K}) = 328$.
123	58	Pavlek, F., and Rogalla, D.	1966	B	4-273	Pure Al, Al 99.9	99.8673 Al, 0.0710 Fe, 0.0420 Si, 0.0140 Mn, and 0.0017 Cu; similar to the above except electrical resistivity ratio $\rho(273\text{ K})/\rho(4.2\text{ K}) = 55.2$, $\rho(293\text{ K})/\rho(20.4\text{ K}) = 57.1$.
124	58	Pavlek, F., and Rogalla, D.	1966	B	4-273	Al 99.7	99.814 Al, 0.1100 Fe, 0.0580 Si, 0.0100 Zn, 0.0040 Ti, 0.0020 Cu, and 0.0020 Mn; similar to the above except electrical resistivity ratio $\rho(273\text{ K})/\rho(4.2\text{ K}) = 28.3$, $\rho(293\text{ K})/\rho(20.4\text{ K}) = 28.6$.
125	60	Moore, J.P., McElroy, D.L., and Parsons, M.	1966	A	100-360	99.999 Al; RR = 520; cylindrical specimen machined from a stock obtained from Reynolds Aluminum Co.; estimated uncertainty ±0.6%.	
126*	61	Wiser, H.	1966	973		No details given.	99.993 Al; rod obtained from British Aluminum Co.; specimen 2.53 cm in diam. and 20.4 cm long.
127	62	Powell, R.W., Tye, R.P., and Woodman, H.J.	1965	A	313-673		99.993 Al; from British Aluminum Co.; specimen 2.81 cm in diam. and 28.0 cm long; smoothed values from table; longitudinal heat flow apparatus used.
128	62	Powell, R.W., et al.	1965	A	323-873		99.993 Al; from British Aluminum Co.; specimen 2.81 cm in diam. and 28.0 cm long; smoothed values from table; longitudinal heat flow apparatus used.
129	62	Powell, R.W., et al.	1965	A	123-323		99.993 Al; from British Aluminum Co.; specimen 8.0 x 0.46 x 0.46 cm; smoothed values from table.
130	63	Powell, R.W., Tye, R.P., and Metcalf, S.C.	1965	A	973-1273		99.993 Al; from British Aluminum Co.; in molten state; smoothed values from table.
131	64	Forewall, K. and Holwech, I.	1964	4.2	Specimen 1		99.99 Al; containing 0.004 Zn; zone refined; bulk resistance ratio $R_{293}/R_{4.2} = 26500$.
132*	64	Forewall, K. and Holwech, I.	1964	4.2	Specimen 2		99.999 Al; containing 0.001 Zn; zone refined; bulk resistance ratio $R_{293}/R_{4.2} = 26500$.
133*	65	Frolo, C. and Dzhitrev, O.	1964	20.4			99.95 Al, 0.05 total impurities; aluminum purified by 15 passages in zone refinement; values measured immediately after deformation in liquid hydrogen; data extracted from figure.
134*	65	Frolo, C. and Dzhitrev, O.	1964	20.4			Similar to above specimen.

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Designation	Composition (weight percent), Specifications and Remarks
135 66	Fenton, E.W., Rogers, J.S., and Woods, S.B.	1963	2-28	Al 3	99.9999 Al; some refined sheet 0.010 in. thick x 0.125 in. diam. rods; supplied by Research Labs. of Consolidated Mining and Smelting Co. of Canada, Trail, British Columbia; acid-etched to remove surface contamination before annealing; rods passed through rollers producing a square cross section that degenerated to rhomboid after several passes; specimen drawn once through steel die to restore cross section to nearly round shape about half way through reduction; further etched to remove surface contamination; annealed in air at 550°C for 10 minutes; $\rho_0 = 0.3903 \times 10^{-8} \Omega \text{ m}$.	
136 66	Fenton, E.W., et al.	1963	2-21	Al 6	Same as the above except $\rho_0 = 0.000268 \times 10^{-8} \Omega \text{ m}$.	
137 67	Parcell, J.R. and Jacobs, R.B.	1963	A	4-30	99.9983 pure	99.9983 Al; specimen (approx.) 0.006 in. x 0.25 in. x 40 in.; supplied by Consolidated Aluminum Co., Jackson, Tennessee; annealed at 350°C for 2 h; $R(300)/R(4) = 1.310$; sample completely immersed in bath of either liquid helium or liquid hydrogen during measurements; resistivities computed from resistance ratios, values used for room temperature resistivity $2.7 \times 10^{-8} \Omega \text{ cm}$ (Butter, J.W. and Reekie, J. [81]); reported error 10%.
138 67	Parcell, J.R. and Jacobs, R.B.	1963	A	4-30	99.999 pure	99.999 Al; approximate specimen dimensions 0.030 in. x 0.125 in. x 40 in.; supplied by A.I.A.G. Metals Inc., New York, New York; annealed at 350°C for 2 h; $R(300)/R(4) = 2.600$; sample completely immersed in bath of either liquid helium or liquid hydrogen during measurements; resistivities computed from resistance ratios, value used for room temperature resistivity $2.7 \times 10^{-8} \Omega \text{ cm}$ (Butter, J.W. and Reekie, J. [81]); reported error 10%.
139* 68	Aleksandrov, B.N. and D'yakov, I.G.	1963	A	273-650	99.9 Al, 0.05 Si, 0.03 Fe; $\rho(23)K = 2.417 \times 10^{-8} \Omega$ assumed; data of Pochapsky [98]; error in resistance $\pm 1\%$.	
140* 68	Aleksandrov, B.N. and D'yakov, I.G.	1963	A	14-290	Single crystal with wire axis coincident with either principal axis or [110] direction; wire diam. 10-15 μ ; data taken from figure.	
141* 68	Aleksandrov, B.N. and D'yakov, I.G.	1963	A	14-261	Polycrystalline Al wire with axis coincident either with the principal axis or with [110] direction; purified by zone melting; $\rho_{12}/\rho_{233} = 3.4 \times 10^{-5}$; below <14 K, $\rho \sim T^3$; data extracted from figure.	
142* 69	Swanson, M.L., Piercy, G.R., and MacKinnon, D.J.	1962	A	1.8	1	99.99 Al; strip specimen 0.003 in. thick; annealed 0.010 in. wires rolled at room temperature; annealed.
143* 69	Swanson, M.L., et al.	1962	A	1.8	2	Similar to the above specimen.
144* 69	Swanson, M.L., et al.	1962	A	1.8	3	Same as above specimen.
145* 69	Swanson, M.L., et al.	1962	A	1.8	4	99.999 Al; strip specimen 0.008 in. thick; annealed 0.010 in. wires rolled at room temperature; annealed.
146* 70	Korol'kov, A.M. and Shestkov, D.P.	1962		294-1073	No details given.	

Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
147*	71	Sirota, H.M.	1962	20-372			No details given; data taken from figure.
148	72	Powell, R.L., Hall, W.J., and Roder, H.M.	1960	A	4-76	Single crystal High purity	99.995 Al, originally single crystal; the JM 360 rod made from Johnson-Matthey stock by Horizons, Inc., Cleveland, Ohio; ground to 3.66 mm diam.; chemical etching after the reduction in diameter indicated the material was still a single crystal; after the last fabrication the rod was annealed in vacuum at about 400°C for 2 h; date extracted from smooth curve; reported error 2%.
149	73	Hedgecock, F.J., Mair, W.B., and Wallingford, E.	1960	A	2.7-26	GRP	<0.002 Cu, <0.002 Fe, <0.002 Mg, <0.001 Mn, <0.001 Si; prepared by the Aluminum Co. of Canada; cold-rolled; annealed in helium at 300°C for 24 h; values calculated from graphically reported ρ/ρ_{300} values using $\rho_{300} = 2.77 \times 10^{-8} \Omega \text{ m}$; reported error 0.5%.
150	74	Simoneas, R.O. and Balluffi, R.W.	1960	287-928	High purity Al	99.995 Al, 0.003 Cu, 0.001 Fe, and 0.001 Si; material donated by Aluminum Co. of America; annealed a few degrees below 933 K for several days, swaged and drawn into 0.43 mm diam. wire; $R(273 \text{ K})/R(4.2 \text{ K}) = 414$ after annealing and essentially the same value for the starting material; resistance ratios corrected for thermal expansion from crude dimensional measurements on specimen $\rho(20^\circ\text{C}) = 2.70 \pm 0.12 \mu\Omega \text{ cm}$; therefore, standard value of $\rho(20^\circ\text{C}) = 2.6548 \mu\Omega \text{ cm}$.	
151	75	DeSorbo, W.	1958	A	1-20	Zone refined	Spectroscopic composition: "trace" of Cu, specimen 0.020 in. diam. x 7-9 ft. long; obtained from W. E. Trogart; single crystal obtained after 6 passes of zone-refining, machined, swaged, and then drawn; between each swaging and each drawing, metal pickled in warm 15% NaOH solution; drawing done with diamond die; heat treatment: annealed for several hours at 550°C and cooled 2-3°C/min.
152	75	DeSorbo, W.	1958	A	1-20	Zone refined	Same sample as above except heat treatment air quenched from 350°C.
153	75	DeSorbo, W.	1958	A	1-20	Zone refined	Same sample as above except heat treatment air quenched from 550°C.
154	75	DeSorbo, W.	1958	A	1-20	Zone refined	Same sample as above except heat treatment fast quenched from 510°C.
155	76	Nikrychov, V.E.	1958	K	139-795		Pure polycrystal; data from figure; error 1-1.5%; Kohlrausch method.
156	77	Nikrychov, V.E.	1957	K	338-797		99.99 Al; polycrystalline.
157	78	Holl, A., Morris, H., and Folger, R.	1957	K	933-1473		Pure liquid Al; data is represented by linear equation ρ (in $\mu\Omega \text{ cm}$) = $0.0146 \cdot T(K) + 10.56$.
158*	79	Broom, T.	1952	B	90-373		99.996+ Al; impurities 0.002 Mg, <0.001 Si, <0.0005 Cu, Fe; wire drawn from 0.183 cm to 0.056 cm diam. then annealed at 500°C for 2 h and furnace cooled; Kelvin double bridge method.
159*	80	Andrews, F.A., Webber, R.T., and Spahr, D.A.	1951	A	4-2,273	Al I	99.996+ Al, 0.001 Mg, 0.001 Si, 0.0006 Fe, 0.0004 Cu, and 0.0004 Mn; single crystal rods, 0.15 in. diam. x 4 in. long; from Alcoa; $\rho_0 = 0.00304 \times 10^{-8} \Omega \text{ m}$; Wenner potentiometer; reported error <2%.
							Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Date Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
160* 80	Andrews, F.A., et al.	1951	A	4.2-273	Al II	Similar to the above specimen; $\rho_0 = 0.00365 \times 10^{-6} \Omega \text{ m}$.
161* 80	Andrews, F.A., et al.	1951	A	4.2-273	Al III	99.99% Al, 0.002 Mg, 0.001 Si, and trace Cu, Fe, and Na; polycrystalline; from Johnson and Matthey; rods 0.15 in. diam. x 4 in. long; $\rho_0 = 0.00551 \times 10^{-6} \Omega \text{ m}$.
162* 81	Butter, J.W. and Reekie, J.	1950	20-297	H-S brand	99.99% Al; polycrystalline; rod specimen; from Johnson, Matthey Ltd.; H-S brand; not cold worked.	
163* 81	Butter, J.W. and Reekie, J.	1950	20-297	H-S	Same as the above specimen except percent reduction of area was 17.9%, i.e., cold worked from annealed state by drawing through diamond dies at uniform speed.	
164* 81	Butter, J.W. and Reekie, J.	1950	20-297	H-S	Same as the above specimen except percent reduction of area was 40.4%.	
165* 81	Butter, J.W. and Reekie, J.	1950	20-297	H-S	Same as the above specimen except percent reduction of area was 60.2%.	
166* 81	Butter, J.W. and Reekie, J.	1950	20-297	H-S	Same as the above specimen except percent reduction of area was 83.1%.	
167* 82	Powell, R. and Evans, E.J.	1942	273	99.99 Al; 0.4 cm x 2.5 cm x 12 cm; electrically refined aluminum from Aluminum Industries, A. G. Mehansen, Switzerland; specimen heated up to the annealing temperature and maintained at that temperature from 2-3 weeks, specimen then allowed to cool slowly to room temperature; resistivity was measured at 273 K, specimen was then heated in furnace and previous annealing temperature was continued for about 3 weeks; after cooling the resistivity of each specimen at 273 K was again determined, this process was continued until no change in resistivity at 273 K was found upon further annealing; density 2.71 g/cm ³ .		
168* 82	Powell, R. and Evans, E.J.	1942	273	Same as the above specimen before annealing.		
169* 83	Taylor, C.S., Valley, L.A., Smith, D.H., and Edwards, J.D.	1938	293	High purity	99.9960 Al (by difference), 0.0020 Si, 0.0010 Cu, 0.0003 Ca, 0.0003 Mg, 0.0003 Na, and 0.0001 Fe; specimen 14 gage sheet, 1 in. wide, 24 in. long; produced by Compagnie des Produits Chimiques et Electrometallurgiques d'Alais Froges et Camargue; electrolytically refined notch-bar ingot remelted in graphite crucible, cast in sheet ingot 1.5 in. thick, cold-rolled to 1 in. thick, surface of slab removed by machining, and further cold-rolled.	
170* 84	Zucker, A. and Warrerup, H.	1935	273.2	99.7 Al.		
171* 85	Kapitza, P.	1929	A	88	Al ₁	99.951 Al, 0.021 Cu, 0.013 Si, 0.012 Fe, 0.002 Ti, 0.001 Vn; wire specimen 0.17 mm in diam. from American Aluminum Co.; resistance ratio $R(290 \text{ K})/R(91 \text{ K}) = 8.77$; units not explicitly given, presume they are in $\Omega \text{ cm}$.

^aNot shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
172*	85	Kapitza, P.	1929	A	88	Al _{II}	Spectroscopic comparison with AlI showed AlII somewhat more impure than AlI; chief impurity copper; strip specimen 0.1 mm thick and about 0.5 mm wide; from Aluminum Co. of America, gift of Dr. Chadwick; resistance ratio R(290 K)/R(91 K) = 7.09; units not explicitly given, presume they are in Ω cm.
173*	85	Kapitza, P.	1929	A	88	Al _{III}	Spectroscopic comparison showed AlIII somewhat more impure than AlII; copper chief impurity; wire specimen 0.15 mm in diam.; from Hartmann and Braun; resistance ratio R(290 K)/R(91 K) = 7.16; units not explicitly given, presume they are in Ω cm.
174*	85	Kapitza, P.	1929	A	88	Al _{III}	The above specimen after magnetoresistivity measurements performed with magnetic field perpendicular to current; resistance ratio R(290 K)/R(80 K) = 8.26; units not explicitly given, presume they are in Ω cm.
175*	86	Stahler, J.	1929		89-476	Pure.	
176*	87	Grüneisen, E. and Coens, E.	1927	A	21.2-273.2	Aluminum 1	Rather pure; source Aluminum Co. of America; turned into small rod from coarse-grained casting; annealed in vacuum at 300°C for 2.5 h; thermal resistivity 0.0500 and 0.289 W cm ⁻² K ⁻¹ at 21.2 and 83.2 K, respectively; Wiedemann-Franz-Lorenz number 1.77 and 1.27 × 10 ⁻⁸ Ω W K ⁻² at 21.2 and 83.2 K, respectively.
177*	87	Grüneisen, E. and Coens, E.	1927	A	21.2-273.2	Al 3	Same as above; grain size 5-15 mm long; drawn and annealed, then stretched 2.57, and recrystallized by annealing thermal resistivity 0.0840 and 0.290 W cm ⁻² K ⁻¹ at 21.2 and 83.2 K, respectively; Wiedemann-Franz-Lorenz number 1.97 and 1.32 × 10 ⁻⁸ V K ⁻² at 21.2 and 83.2 K, respectively.
178*	87	Grüneisen, E. and Coens, E.	1927	A	21.2-273.2	Al 100	Technically pure; source unknown, commercial conductor; annealed in vacuum at 250°C; thermal resistivity 0.341 and 0.374 W cm ⁻² K ⁻¹ at 21.2 and 83.2 K, respectively; Wiedemann-Franz-Lorenz number 2.18 and 1.47 × 10 ⁻⁸ Ω W K ⁻² at 21.2 and 83.2 K, respectively.
179*	87	Grüneisen, E. and Coens, E.	1927	A	21.2-273.2	Al 101	Same as above; after annealing stretched 32 and recrystallized by annealing thermal resistivity 0.470 and 0.408 W cm ⁻² K ⁻¹ at 21.2 and 83.2 K, respectively; Wiedemann-Franz-Lorenz number 2.20 and 1.55 × 10 ⁻⁸ Ω W T ⁻¹ at 21.2 and 83.2 K, respectively; measuring length = 2 crystal grains.
180*	87	Grüneisen, E. and Coens, E.	1927	A	21.2-273.2	Al 21	Moderately pure; single crystal; grown by recrystallization; thermal resistivity 0.730 and 0.481 W cm ⁻² K ⁻¹ at 21.2 and 83.2 K, respectively; Wiedemann-Franz-Lorenz number 2.20 and 1.55 × 10 ⁻⁸ Ω W K ⁻² at 21.2 and 83.2 K, respectively.
181	88	Matuyama, Y.	1927	-	939-1198	Chemically pure; melting point 931.65 K, r = 2.58 mm, t = 62.3 mm, σ _{em} = 25.5 × 10 ⁻⁸ .	

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ALUMINUM-Al (continued)

Data Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Identification	Composition (weight percent), Specifications and Remarks
182* 89	Smith, A.W.	1925	B	296.2		99.97 Al; 1.9 cm diam. x 10 cm long; specimen from Aluminum Co. of America.
183 90	Schofield, F.H.	1925	A	289-814		99.7 Al; free from discontinuities between core and surrounding layers, inclusion of dross, oxidized skin, and unsoundness; supplied by British Aluminum Co., Ltd.; 6.75 in. diam. billets cast from a maximum temperature of 913 K, annealed at 773 K for 2.5 h, extruded at 693 K to 0.75 in. diam.; annealed at 723 K for 2.5 h; density 2.70 g cm ⁻³ at 294 K; reported error 1%.
184* 91	Holborn, L.	1921		273,293	Al IV	99.59 Al; 0.22 Si, 0.18 Fe, and 0.01 C.
185* 91	Holborn, L.	1921		273,293	Al IV	Same as the above except specimen was annealed.
186* 91	Holborn, L.	1921		273,293	Al VI	99.9 Al, 0.06 Cu, 0.02 Si, and trace of Fe; wire specimen 1 mm in diam. and 7.3 m wound on porcelain tube; material from specimen Al IV above purified, drawn by Heraeus.
187* 91	Holborn, L.	1921		273,293	Al VI	Above specimen annealed for a long time at 250°C.
188* 92	Holborn, L.	1919		20-195		99.6 Al, 0.4% impurities; polycrystalline.
189 93	Bornmann, K. and Hagemann, K.	1914		973-1573		Pure aluminum specimen was obtained from Neuhausen
190* 94	Wolff, F.A. and Dallinger, J.H.	1911		293		99.52-99.60 Al, 0.26-0.36 Si, and 0.14-0.15 Fe; commercial hard-drawn aluminum wire; density 2.70 g cm ⁻³ .
191 95	Niccolai, G.	1908		64-673		Wire specimen obtained from Firma C.A.F. Kahlbaum; 0.5 mm diam. x 8 m long.

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al
[Temperature, T; K; Electrical Resistivity, ρ , 10^{-8} $\Omega \text{ m}$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 1*</u>															
1.600	0.93577×10^{-8}	2.631	0.69215×10^{-8}	3.770	1.3720×10^{-8}	3.593	2.9857×10^{-8}	1.294	110.640×10^{-8}	2.166	22.461				
1.650	0.93631	3.796	0.73259	3.978	1.3836×10^{-8}	3.785	2.9967×10^{-8}	1.535	110.642×10^{-8}	2.620	22.470				
1.700	0.93686	4.221	0.75419×10^{-8}	4.132	1.3929×10^{-8}	3.978	3.0090×10^{-8}	1.774	110.645×10^{-8}	3.016	22.481				
1.755	0.93750			4.220	1.3985×10^{-8}	4.133	3.0198×10^{-8}	1.990	110.648×10^{-8}	3.402	22.496				
1.800	0.93804							2.167	110.651×10^{-8}	3.699	22.509				
1.850	0.93867	2.362	0.68626×10^{-8}	1.950	1.3135×10^{-8}	1.301	2.9268×10^{-8}	2.611	110.661×10^{-8}	3.945	22.523				
1.904	0.93937							3.011	110.675×10^{-8}	4.211	22.540×10^{-8}				
1.950	0.94001	2.582	0.69098	2.000	1.3143×10^{-8}	1.700	2.9290×10^{-8}	3.400	110.679×10^{-8}						
2.000	0.94073	2.728	0.69456	2.050	1.3151×10^{-8}	1.800	2.9303×10^{-8}	3.800	110.721×10^{-8}						
2.050	0.94148	2.989	0.70186	2.329	1.3200×10^{-8}	2.171	2.9362×10^{-8}	4.200	110.755×10^{-8}						
2.100	0.94226	3.1875	0.70619	2.101	1.3160×10^{-8}	1.900	2.9317×10^{-8}			1.5	15.4×10^{-8}				
2.145	0.94299	3.378	0.71501	2.171	1.3172×10^{-8}	2.000	2.9332×10^{-8}			4.1	15.4				
2.156	0.94318	3.596	0.72343	2.329	1.3200×10^{-8}	2.171	2.9362×10^{-8}			6.2	15.9				
2.167	0.94336	3.797	0.73302	2.468	1.3232×10^{-8}	2.360	2.9402×10^{-8}			9.0	17.0				
2.170	0.94341	3.997	0.74195×10^{-8}	2.80	1.3323×10^{-8}	2.578	2.9452×10^{-8}			9.7	18.1				
2.171	0.94343×10^{-8}									11.6	19.2				
<u>DATA SET 6*</u>															
1.298	0.93306×10^{-8}	4.134	0.74837×10^{-8}	1.292	1.3060×10^{-8}	4.221	3.0264×10^{-8}			3.019	6.6282×10^{-8}				
1.302	0.93308	4.224	0.7538804×10^{-8}	1.402	1.3069×10^{-8}					3.407	6.6288×10^{-8}				
1.322	0.93324									3.019	6.6298×10^{-8}				
1.355	0.93351									3.407	6.6314×10^{-8}				
1.362	0.93356	1.180	0.67167×10^{-8}	1.601	1.3090×10^{-8}	4.209	3.0254×10^{-8}			3.773	6.6332×10^{-8}				
1.363	0.93358	1.191	0.67154	1.450	1.3095×10^{-8}	4.221	3.0264×10^{-8}			4.167	6.6361×10^{-8}				
1.402	0.93190	1.225	0.67176	1.701	1.3101×10^{-8}					4.206	6.6388×10^{-8}				
1.453	0.93436	1.298	0.67231	1.750	1.3107×10^{-8}					4.628	6.6481×10^{-8}				
1.500	0.93479	1.401	0.67316	1.850	1.3114×10^{-8}	1.254	2.9246×10^{-8}			4.793	6.6594×10^{-8}				
1.550	0.93526	1.500	0.67406	1.900	1.3128×10^{-8}	1.353	2.9254×10^{-8}			4.101	6.6693×10^{-8}				
2.171	0.94341	1.601	0.67506	1.701	1.3107×10^{-8}	1.402	2.9259×10^{-8}			4.206	6.7198×10^{-8}				
2.647	0.93304	1.701	0.67616	1.541	1.3114×10^{-8}	1.541	2.9272×10^{-8}			4.206	6.7198×10^{-8}				
2.647	0.93597	1.801	0.67737	1.601	1.3114×10^{-8}	1.601	2.9279×10^{-8}			2.102	6.0176×10^{-8}				
2.905	0.93982	1.901	0.67868	1.253	1.3056×10^{-8}	1.300	2.9250×10^{-8}			2.613	6.0277×10^{-8}				
3.149	0.96149	2.001	0.68024	1.289	1.3059×10^{-8}					3.022	6.0392×10^{-8}				
3.401	0.97660	2.101	0.68168	1.352	1.3065×10^{-8}	1.353	2.9254×10^{-8}			3.395	6.0534×10^{-8}				
3.646	0.98661	2.112	0.68286×10^{-8}	1.451	1.3074×10^{-8}	1.522	106.801×10^{-8}			3.803	6.0737×10^{-8}				
3.842	0.99557									4.193	6.0981×10^{-8}				
4.042	1.00356									4.229	6.0995×10^{-8}				
4.106	1.00922×10^{-8}										5.9	1.12			
<u>DATA SET 2*</u>															
1.600	0.93577×10^{-8}	2.578	1.1260×10^{-8}	2.049	2.9361×10^{-8}	2.610	106.821×10^{-8}			8.2	1.68				
2.171	0.94341×10^{-8}	2.986	1.3300	2.100	2.9349×10^{-8}	2.994	106.834×10^{-8}			10.1	1.70				
3.186	1.3382									3.389	106.853×10^{-8}				
3.318	1.3454									3.189	106.852×10^{-8}				
3.318	1.3532									3.80	106.879×10^{-8}				
3.596	1.3632									3.183	106.916×10^{-8}				
<u>DATA SET 13*</u>															
1.600	0.93577×10^{-8}	2.578	1.1260×10^{-8}	2.049	2.9361×10^{-8}	2.610	106.821×10^{-8}			8.2	1.68				
2.171	0.94341×10^{-8}	2.986	1.3300	2.100	2.9349×10^{-8}	2.994	106.834×10^{-8}			10.1	1.70				
3.186	1.3382									3.389	106.853×10^{-8}				
3.318	1.3454									3.189	106.852×10^{-8}				
3.318	1.3532									3.80	106.879×10^{-8}				
3.596	1.3632									3.183	106.916×10^{-8}				

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
<u>DATA SET 23 (cont.)*</u>													
		<u>DATA SET 25 (cont.)*</u>		<u>DATA SET 27 (cont.)*</u>		<u>DATA SET 30*</u>		<u>DATA SET 36 (cont.)</u>		<u>DATA SET 41 (cont.)</u>			
20.7	9.24	12.1	7.55	4.1	12.7	4.2	5.517 $\times 10^{-8}$	20	53.5	17	154.0		
23.1	12.9	16.2	9.17	5.7	12.7			25	66.0	20	160.0		
25.0	18.3	16.3	10.2	8.0	13.8			32.8	118.0 $\times 10^{-8}$	25	176.0		
26.7	26.1	17.9	12.4	9.7	14.9	4.2	6.706 $\times 10^{-8}$	<u>DATA SET 37</u>	32.8	47.9	555.0 $\times 10^{-8}$		
28.1	30.5	20.7	14.5	11.2	15.5								
29.2	37.4	21.8	17.2	13.0	17.1	<u>DATA SET 32</u>	10	50.7 $\times 10^{-8}$	<u>DATA SET 42</u>				
31.1	50.2	23.9	20.4	15.4	19.2								
32.2	61.3	26.0	28.4	17.6	21.0	10	14.3 $\times 10^{-8}$	13	51.9	10	149.0 $\times 10^{-8}$		
33.6	75.7	27.7	34.7	19.3	24.0								
34.7	86.8	29.7	47.0	21.7	28.8	13	15.9	20	58.9	13	151.0		
35.6	98.0	31.7	60.8	23.9	34.7	17	19.1	25	72.5	17	156.0		
37.1	121.0	33.1	74.1	25.7	40.0	20	20.6	32.8	126.0 $\times 10^{-8}$	20	161.0		
39.3	155.0	34.9	91.6	27.5	48.5	25	23.7			25	179.0		
41.1	195.0	36.7	116.0	29.9	60.2	32.8	85.9 $\times 10^{-8}$	<u>DATA SET 38</u>	32.8	47.9	241.0		
42.0	217.0 $\times 10^{-8}$	39.0	151.0	31.7	73.5								
		40.4	182.0 $\times 10^{-8}$	32.6	82.0	<u>DATA SET 33</u>	10	89.4 $\times 10^{-8}$	<u>DATA SET 43</u>				
<u>DATA SET 26*</u>													
1.5	3.73 $\times 10^{-8}$	<u>DATA SET 26*</u>		34.7	102.0	10	22.9 $\times 10^{-8}$	13	92.5	<u>DATA SET 43</u>			
4.4	3.23	1.5	9.56 $\times 10^{-8}$	36.0	119.0	13	24.1	17	98.2				
5.5	3.77	4.3	9.59	39.5	139.0	20	105.0	13	92.1 $\times 10^{-8}$				
7.3	4.32	5.7	10.6	41.5	176.0	17	27.2	25	123.0	20	107.0 $\times 10^{-8}$		
10.0	4.35	6.6	10.6	<u>DATA SET 28</u>	218.0 $\times 10^{-8}$	20	31.0	32.8					
12.2	4.90	8.0	10.7		25	43.6	47.9	500.0 $\times 10^{-8}$	<u>DATA SET 44</u>				
13.9	5.45	9.0	11.2	1.5	1.02 $\times 10^{-8}$	<u>DATA SET 34</u>	32.8	95.4 $\times 10^{-8}$		10	98.0 $\times 10^{-8}$		
15.8	6.53	10.3	11.7	4.2	1.08				13	100.0			
17.9	8.68	13.0	13.4	6.0	1.27	10	31.5 $\times 10^{-8}$	13	122.0 $\times 10^{-8}$	17	106.0		
19.6	10.2	14.6	15.0	7.5	1.49				124.0	20	112.0		
21.5	13.5	17.6	17.7	8.5	1.63	13	32.6 $\times 10^{-8}$	17	129.0	25	129.0		
23.5	16.1	19.5	20.9	10.0	1.76	17	35.6	20	136.0	32.8	187.0 $\times 10^{-8}$		
25.4	21.5	21.3	23.5	12.5	2.64	25	39.3	25	151.0				
27.1	27.3	22.8	27.3	15.0	3.70	32.8	102.0 $\times 10^{-8}$	47.9	208.0	<u>DATA SET 45*</u>			
29.3	39.0	26.7	31.5	17.5	5.14				528.0 $\times 10^{-8}$				
30.6	48.0	26.5	38.4	20.0	7.62	<u>DATA SET 40</u>	10	122.0 $\times 10^{-8}$	10	103.0 $\times 10^{-8}$			
32.8	65.6	43.2	27.7	22.5	11.4				13	106.0			
33.1	68.2	29.2	51.2	25.0	17.8	10	39.7 $\times 10^{-8}$	13	128.0 $\times 10^{-8}$	17	111.0		
34.2	79.9 $\times 10^{-8}$	30.3	57.0	27.5	28.5	13	43.6	17	135.0	25	133.0		
		31.7	66.6	30.0	43.4								
		33.9	87.3	32.5	64.6	<u>DATA SET 35</u>	20	47.6	140.0	140.0	141.0		
		36.1	113.0	35.0	92.1			25	157.0				
		37.8	137.0	37.5	130.0	<u>DATA SET 36</u>	32.8	112.0 $\times 10^{-8}$	32.8	214.0			
		41.5	175.4 $\times 10^{-8}$	40.0	175.4 $\times 10^{-8}$			47.9	534.0 $\times 10^{-8}$	10	112.0 $\times 10^{-8}$		
		5.36	41.5	210.0 $\times 10^{-8}$	<u>DATA SET 29</u>	10	45.4 $\times 10^{-8}$	<u>DATA SET 41</u>	17	119.0			
		6.44	6.44	<u>DATA SET 27*</u>	4.2	5.80 $\times 10^{-8}$	13	46.5	25	125.0			
		8.9	6.46					17	49.4	13	141.0		
		10.5	7.54	1.5	12.7 $\times 10^{-8}$				150.0	13	200.0		

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>	<u>T</u>	<u>P</u>
<u>DATA SET 47*</u>													
10	119.0×10^{-8}	5.41	2.238	1.5	4.65×10^{-8}	3.69	13.61	15.8	18.2	7.66	3.10		
13	121.0	5.52	2.244	2.0	4.66	4.02	13.63	17.8	20.7	10.0	3.71		
17	125.0	5.61	2.255	3.0	4.68	13.64	4.26	19.8	23.9	12.1	4.01		
25	148.0	5.72	2.267	4.0	4.71×10^{-8}	4.65	13.66	21.6	28.6	13.8	4.95		
32.8	205.0×10^{-8}	5.91	2.275			5.38	13.75	23.6	35.9	17.8	8.09		
<u>DATA SET 48*</u>													
1.35	2.068×10^{-8}	6.12	2.294	5.92	2.284		6.23	13.77	25.5	42.3	19.6	10.6	
1.46	2.069	6.12	2.301	9.33	24.2		6.41	13.85	27.6	51.8	21.6	13.7	
1.55	2.071	6.12	2.316				7.20	13.92	29.6	65.4	23.7	19.1	
1.64	2.073	6.16	2.323×10^{-8}				7.63	14.03	31.2	79.0×10^{-8}	25.7	25.4	
1.79	2.077						8.33	14.09			27.6	34.6	
1.86	2.079						8.83	14.19			29.5	47.6	
1.95	2.080						9.09	14.28×10^{-8}			31.3	61.3×10^{-8}	
2.07	2.081												
2.18	2.083												
2.29	2.085												
2.41	2.089												
2.50	2.091												
2.58	2.094												
2.69	2.097												
2.81	2.099												
2.92	2.102												
2.98	2.104												
3.10	2.108												
3.22	2.112												
3.32	2.116												
3.38	2.122												
3.50	2.124												
3.61	2.130												
3.72	2.134												
3.84	2.137												
3.95	2.139												
4.06	2.143												
4.13	2.152												
4.24	2.159												
4.35	2.167												
4.47	2.171												
4.55	2.177												
4.70	2.182												
4.75	2.188												
4.87	2.198												
4.96	2.204												
5.10	2.208												
5.18	2.215												
5.30	2.223												

*Not shown in figure.

DATA SET 54*DATA SET 55*DATA SET 56*DATA SET 57*DATA SET 58*DATA SET 59*DATA SET 60*DATA SET 61*DATA SET 62*DATA SET 63*DATA SET 64*DATA SET 65*DATA SET 66*DATA SET 67*DATA SET 68*DATA SET 69*DATA SET 70*DATA SET 71*DATA SET 72*DATA SET 73*DATA SET 74*DATA SET 75*DATA SET 76*DATA SET 77*DATA SET 78*DATA SET 79*DATA SET 80*DATA SET 81*DATA SET 82*DATA SET 83*DATA SET 84*DATA SET 85*DATA SET 86*DATA SET 87*DATA SET 88*DATA SET 89*DATA SET 90*DATA SET 91*DATA SET 92*DATA SET 93*DATA SET 94*DATA SET 95*DATA SET 96*DATA SET 97*DATA SET 98*DATA SET 99*

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM^a (continued)

WATER SUPPLY IN INDIA

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM Al (continued)

T	p	T	p	T	p	T	p	T	p	T	p	T	p
<u>DATA SET 104</u>		<u>DATA SET 109</u>		<u>DATA SET 113 (cont.)*</u>		<u>DATA SET 116</u>		<u>DATA SET 121*</u>		<u>DATA SET 126*</u>			
300	2.65	321	3.03	20.4	67.1×10^{-8}	883	10.3	4.2	0.001111	973	25		
400	3.45	340	3.09	21.8	69.5	922	10.8	20.4	0.002388				
500	4.90	367	3.57	23.0	72.1	933	20.1	77	0.221				
600	6.05	397	3.85	24.4	75.1	936	20.2	195	1.44				
700	7.15	426	4.12	25.4	78.3	939	20.6	273	2.46	313	2.86		
800	8.30	441	4.32	26.4	82.3	943	20.7			373	3.56		
900	9.40	496	4.94	27.8	88.7	944	20.4	473	4.73				
		605	6.24	28.9	95.0	947	20.5	573	5.90				
<u>DATA SET 105</u>		655	6.79	30.2	102.6	960	20.6	4.2	0.00575	673	7.12		
10	0.00075	662	6.92	31.4	110.8	963	20.7	20.4	0.00766				
20	0.00143	693	7.33	32.3	118.1	974	20.7	77	0.224				
30	0.00532	<u>DATA SET 110</u>		33.2	126.8	998	21.0	195	1.50				
40	0.01867	77	0.221	34.1	136.8	1001	21.0	273	2.30	323	2.98		
50	0.04907	273	2.425	<u>DATA SET 114</u>		1004	21.1			373	3.56		
60	0.09640			6.7	22.7×10^{-8}	1006	21.1	473	4.73				
70	0.1632			<u>DATA SET 111*</u>		1008	21.2	573	5.90				
80	0.2655			9.4	23.5	1014	21.2	673	7.12				
90	0.3400			9.4	23.5	1025	21.4	20.4	0.0439				
100	0.4425	4.6	0.00261	11.4	24.5	1034	21.3	77	0.0463	773	8.31		
120	0.66331	15.1	0.00383	13.3	25.6	1044	21.7	195	1.62	673	9.92		
140	0.8934	17.7	0.00380	16.1	27.7	1050	21.6						
160	1.137	21.0	0.00109	18.5	30.2	1078	22.1						
180	1.361	26.9	0.00505	20.1	32.2	1080	22.2						
200	1.593	27.8	0.00768	21.8	34.7	<u>DATA SET 117</u>							
220	1.824	30.1	0.0103	23.3	37.8	4.2	0.00725	273	2.42				
240	2.053	35.4	0.0156	24.7	41.1			77	0.341				
260	2.280	41.9	0.0302	25.8	43.9	<u>DATA SET 118</u>		195	1.63				
273.2	2.430	48.1	0.0461	26.9	48.1			273	2.46				
295	2.618	54.0	0.0647	27.9	52.4	<u>DATA SET 125</u>				973	26.3		
		61.3	0.103	29.0	57.2	298	2.74			1023	27.1		
<u>DATA SET 106*</u>		69.8	0.169	29.9	61.9	373	3.57			1073	27.8		
273.2	2.485	80.0	0.260	30.6	66.4	473	4.71	100	0.441	1123	28.6		
351.2	3.33	90.3	0.375	31.6	73.4	573	5.86	120	0.668	1173	29.3		
408.4	4.05			32.6	81.6	673	7.06	140	0.901	1223	30.1		
<u>DATA SET 107</u>		<u>DATA SET 112*</u>		33.4	88.6	773	8.30	160	1.133	180	1.367	1273	30.9
193.7	1.52	1.5	0.00428	34.3	100.6×10^{-8}	<u>DATA SET 119</u>		200	1.599				
248.2	2.14	293	2.84	<u>DATA SET 115</u>		20.4	40.9×10^{-8}	220	2.062				
298.2	2.72			<u>DATA SET 113*</u>		<u>DATA SET 120*</u>		260	2.292				
351.2	3.33			77.78	0.2257			280	2.322				
408.4	4.05			19.6	1.554			300	2.751				
<u>DATA SET 108*</u>		11.2	57.7×10^{-8}	213.2	2.430	<u>DATA SET 121*</u>		320	2.982				
293	2.9	13.2	59.7	298.5	2.724	20.4	69.7×10^{-8}	340	3.211				
		15.8	61.7					360	3.443				
		18.9	65.0										

not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM A1 (continued)

<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>	<u>T</u>	<u>p</u>
<u>DATA SET 133*</u>		<u>DATA SET 139*</u>		<u>DATA SET 142*</u>		<u>DATA SET 146 (cont.)</u>		<u>DATA SET 150 (cont.)</u>		<u>DATA SET 154 (cont.)</u>			
20.4	0.698	21.3	2.417	1.8	37.6×10^{-8}	47.00	0.08630	895.0	10.117	17.9	106.3		
		37.3	2.500			76.04	0.2301	905.6	10.292	16.8	108.8		
<u>DATA SET 134*</u>		47.3	4.619	<u>DATA SET 143*</u>				915.9	10.449	19.7	111.4×10^{-8}		
20.4	0.624	57.3	5.801	1.8	20.0×10^{-8}		<u>DATA SET 149</u>	927.8	10.644			<u>DATA SET 155</u>	
<u>DATA SET 135</u>		67.3	7.046										
		77.3	8.346	<u>DATA SET 144*</u>		2.7	80.6×10^{-8}						
		82.3	9.170			4.1	80.6						
		87.3	9.755			5.3	80.9						
2.24	9.0×10^{-8}	89.8	10.139	1.8	17.5×10^{-8}	5.7	80.9	1.5	29.2×10^{-8}	40.6	4.11		
4.00	9.1	92.3	10.540			6.3	82.0	2.3		45.9	4.82		
6.40	9.4			<u>DATA SET 145*</u>		6.7	82.0	3.0		55.2	6.04		
6.40	9.5					7.2	82.8	4.0		63.4	7.01		
8.55	10.0			1.8	14.9×10^{-8}	7.8	83.9	14.1		72.8	8.36		
9.35	10.2					8.3	83.9	16.0		79.5	9.34		
10.2	10.4			<u>DATA SET 146*</u>		8.7	82.3	18.0					
12.51	12.3					9.2	81.7	20.0	44.1×10^{-8}				
15.3	15.0	14.2	0.000153	294	2.8	10.3	85.6					<u>DATA SET 156</u>	
19.6	22.2	15.5	0.000211	372	3.8	11.0	83.7						
23.4	40.4	16.2	0.000266	492	5.4	11.7	83.9						
28.0	96.3×10^{-8}	17.0	0.000334	618	6.7	12.6	84.5	1.5	33.6×10^{-8}	40.6	3.15		
		18.6	0.000383	721	7.5	13.4	87.5	2.3		553.0	5.85		
<u>DATA SET 136</u>		20.3	0.000482	822	9.0	14.5	87.5	3.0		631.0	6.82		
		21.2	0.000607	914	9.7	16.1	91.4	4.0		730.2	8.20		
2.19	5.70×10^{-8}	59.5	0.0825	942	25.5	19.1	99.4	14.2		796.6	9.14		
2.56	5.70	68.1	0.119	976	25.9	21.9	106.0	16.0					
3.06	5.70	85.3	0.227	1024	26.1	26.1	126.0	10^{-8}					
3.59	5.72	97.7	0.328	1073	27.7								
3.91	5.77	128.3	0.569										
7.31	6.31	290.4	2.16	<u>DATA SET 147*</u>		287.4	2.586						
8.94	6.98					289.4	2.609	1.5	55.4×10^{-8}	117.3	22.77		
9.56	7.07			<u>DATA SET 141*</u>		293.0	2.653	2.3		127.3	29.24		
13.68	9.82					296.1	2.687	3.0		137.3	30.71		
16.38	13.3	14.2	0.000158	186.6	1.252	377.4	3.605	4.0		147.3	32.17		
17.10	14.4	14.8	0.000218	272.6	2.362	387.8	3.726	14.3					
21.39	26.5×10^{-8}	16.3	0.000286	372.0	3.445	438.4	4.304	16.0					
		17.8	0.000343			502.0	5.041	18.0					
<u>DATA SET 137</u>		17.8	0.000376	<u>DATA SET 148</u>		583.6	6.006	19.9	73.5×10^{-8}	90.2	0.352		
		19.5	0.000494			688.3	7.286			194.7	1.55		
4	20.0×10^{-8}	20.4	0.000649	4.00	0.02488	728.8	7.802			273.2	2.44		
20	28.2	58.0	0.000809	7.33	0.02489	764.4	8.264			373.2	3.59		
30	69.3×10^{-8}	60.7	0.116	13.40	0.02606	770.9	8.355	2.0	91.6×10^{-8}			<u>DATA SET 158*</u>	
<u>DATA SET 138</u>		67.4	0.347	19.77	0.02805	811.3	9.185	3.0					
		115	0.627	21.98	0.02992	843.3	9.345	4.2					
4	0.00106	261	2.25	25.01	0.03404	854.3	9.516	13.9		4.2	0.00304		
20	0.00201			28.91	0.04018	872.5	9.774	14.9		100.0	102.0		
30	0.00514			35.08	0.05346	878.1	9.868	16.9		103.9			

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ALUMINUM (continued)

Not shown in figure.

3.2. Manganese

There are 16 references available reporting temperature dependence of the electrical resistivity from 1 to 1873 K. However, the data are highly contradictory, and in several cases disagree both qualitatively and quantitatively. Further careful measurements on purer samples covering the entire temperature range, especially above 300 K and below 20 K, are required and strongly recommended. The information on specimen characterization and on measurement condition for each of the data sets is given in Table 5. The data sets are tabulated in Table 6 and partially shown in Figs. 4 and 5.

Electrical resistivity data on polycrystalline manganese reported earlier are much higher than those reported recently. These differences may be possibly due to the low purity and insufficient heat treatment of the manganese samples studied earlier. Meaden and Pelloux-Gervais³⁰² demonstrated that the room-temperature electrical resistivity dropped from $205 \times 10^{-8} \Omega \text{ m}$ to $144.2 \times 10^{-8} \Omega \text{ m}$ after annealing the specimen at 898 K.

Meaden³⁰³ (data set 10), Bellou and Coles³⁰⁶ (data set 14), and White and Woods³⁰⁷ (data set 15), have reported T^2 dependence of the temperature-dependent resistivity (ρ_i) below 17 K. This was confirmed by Nagasawa and Senba³⁰⁰ (data set 4) and by Murayama and Nagasawa³¹⁰ (data set 19). The recommended values from 20-325 K are based on the generally agreed upon data of Nagasawa and Senba³⁰⁰ (data set 4), Meaden and Pelloux-Gervais³⁰², (data set 12), Bellou and Coles³⁰⁶ (data set 14), and of White and Woods³⁰⁷ (data set 16). The recommended values below 20 K for $\rho_0 = 6.9 \times 10^{-8} \Omega \text{ m}$ are based on the data of Meaden³⁰³ (data set 10) and Meaden and Pelloux-Gervais³⁰⁴ (data set 12).

An appreciable spin-disorder contribution is indicated by large resistivity values. It appears that the spin-disorder contribution generally present at higher temperatures still remains at liquid helium temperatures. The temperature dependent resistivity (ρ_i) falls linearly and slowly with temperature below 325 K. It goes through a minimum at about 94 K, and then remains practically constant for 4 to 5 degrees before increasing to a weak maximum at 70 K. Below this temperature, ρ_i drops very rapidly, finally becoming proportional to T^2 below 17 K.

Alpha-Mn is a stable phase below 980 K and has a complex cubic (A12) crystal structure with 58 atoms in the unit cell. At 980 K, α -Mn transforms to β -Mn which has a complex cubic structure (A13) with 20 atoms in the unit cell. It is possible to retain the β phase at room temperature by rapid quenching from 980–1300 K. Brunke³¹¹ obtained a value of $91 \times 10^{-8} \Omega \text{ m}$ for the electrical resistivity of β -Mn. Potter et al.³¹² and Erfling³¹³ have reported about $40 \times 10^{-8} \Omega \text{ m}$ for the room-temperature electrical resistivity of fct γ -Mn. High-temperature δ -Mn with a bcc structure is stable between 1411 and 1519 K.

There are only two data sources available in the temperature range 325–1519 K. Grube and Speidel³⁰⁸ (data set 17) reported that the resistivity of manganese increases slowly with increasing temperature from 325 to 980 K and then decreases sharply from 980 to 1519 K. However, Akshentsev et al.³⁰¹ (data sets 5,6) reported that the electrical resistivity rises sharply between 800–980 K, then slowly from 980 to 1300 K followed by a slow decrease from 1300 to 1400 K and then further increases. The reliability of these results is questionable. Room-temperature electrical resistivity of Grube and Speidel³⁰⁸ (data set 17) is twice as much as the recommended value, and indicates a high impurity in their sample. The value of $38 \times 10^{-8} \Omega \text{ m}$ at 800 K for the electrical resistivity reported by Akshentsev et al.³⁰¹ (data set 5) is far lower than the recommended room-temperature value of $144 \times 10^{-8} \Omega \text{ m}$. Therefore, these data are rejected. The recommended values from 325 to 700 K are obtained by extrapolating the low-temperature data.

The published work on the electrical resistivity of molten manganese is equally contradictory. For instance, Akshentsev et al.³⁰¹ (data set 6) reported an increase in the resistivity with temperature, contrary to the results of Levin et al.²⁹⁸ (data set 2) and of Vostryakov et al.³⁰⁵ (data set 13) who reported a decrease in the resistivity with temperature. On the other hand, Grube and Speidel³⁰⁸ (data set 17) reported a constant value of $40 \times 10^{-8} \Omega \text{ m}$ from 1523 to 1543 K. Summarizing this, the electrical resistivity at the melting point varies from 40 to $190 \times 10^{-8} \Omega \text{ m}$. Therefore, the available data and information at and above melting point cannot be used for meaningful data analysis. Consequently, no recommendations were made for the electrical resistivity of manganese in the melting region.

The recommended values of the electrical resistivity given in Table 3 and shown in Figs. 4 and 5 along with the experimental data are for manganese of

purity 99.99% or higher, but those below room temperature are applicable specifically to manganese with $\rho_0 = 6.90 \times 10^{-8} \Omega \text{ m}$. The table gives both values uncorrected and corrected for thermal expansion, while the figure shows only the uncorrected values. The thermal expansion values needed for such correction are taken from ref. 314. The uncertainty in the recommended values is estimated to be within $\pm 10\%$ from 7 to 100 K and above 300 K, and $\pm 5\%$ below 7 K and from 100 to 300 K.

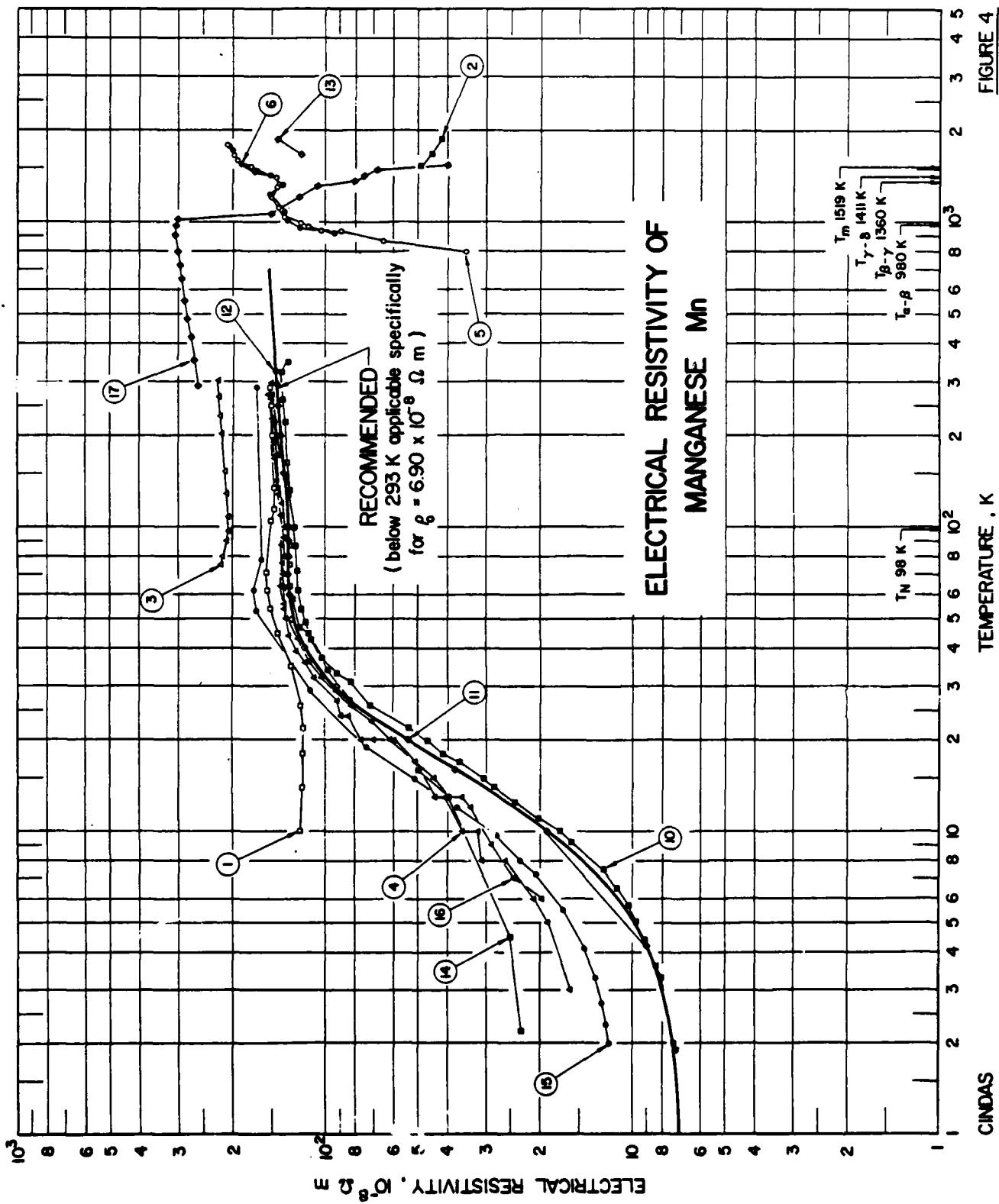
The effect of a magnetic field on the resistivity of manganese at low temperature is relatively small compared with that for pure copper. Meaden³⁰³ found that a magnetic field of 18.5 kOe increases the resistivity by 10.5% at 4.2 K, 9% at 5.4 K, 8% at 5.9 K, and 0.2% at 77 K. Murayama and Nagasawa³¹⁰ (data set 19) studied temperature and magnetic field dependence of the resistivity of polycrystalline α -Mn and observed that the anomalously large coefficient of T^2 term in the low temperature resistivity decreased appreciably for an increase in the applied field, suggesting the suppression of spin fluctuations in the antiferromagnetic α -Mn by the high applied field. Those readers seeking additional information on the effect of magnetic field on the electrical resistivity of manganese are directed to refs. 315-341.

Adanu and Grassie²⁹⁷ (data set 1) studied the temperature dependence of the electrical resistivity of a thin manganese film. For a film of thickness 4000 Å formed on a thin glass substrate, they found that the resistivity decreased linearly as the temperature was reduced from room-temperature, then passed through a minimum at ~ 120 K and a maximum at ~ 70 K, followed by a sharp drop before going through another minimum at 22 K. These features of the resistivity of thin films, with the exception of the minimum at ~ 22 K, are qualitatively similar to those reported for bulk specimens reported by Meaden and Pelloux-Gervais³⁰² (data set 12) and by White and Woods³⁰⁷ (data sets 15,16). Additional information/data on films are reported in refs. 342-350. The pressure dependence of the electrical resistivity is reported in refs. 352-355.

TABLE 4. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF MANGANESE^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{ m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
0	6.90	6.88	94	131.9	131.4
1	7.02	7.00	100	132.5	132.1
4	8.82	8.79	150	136.3	135.9
7	12.78	12.74	200	139.4	139.1
10	18.90	18.84	250	142.0	141.9
15	33.9	33.8	273	143.1	143.0
20	53.8	53.6	293	144.0	144.0
25	75.8	75.6	300	144.2	144.2
30	93.7	93.4	350	145.9	146.1
40	116.0	115.6	400	147.3	147.7
50	126.5	126.1	500	149.4	150.1
60	131.2	130.7	600	150.9	152.1
70	133.0	132.5	700	151.9	153.6
80	132.5	132.0			
90	132	131.5			

^aThe values are for well-annealed manganese of purity 99.99% or higher, but those below room temperature are applicable specifically to manganese having a residual resistivity of $6.90 \times 10^{-8} \Omega \text{ m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively.

**FIGURE 4**

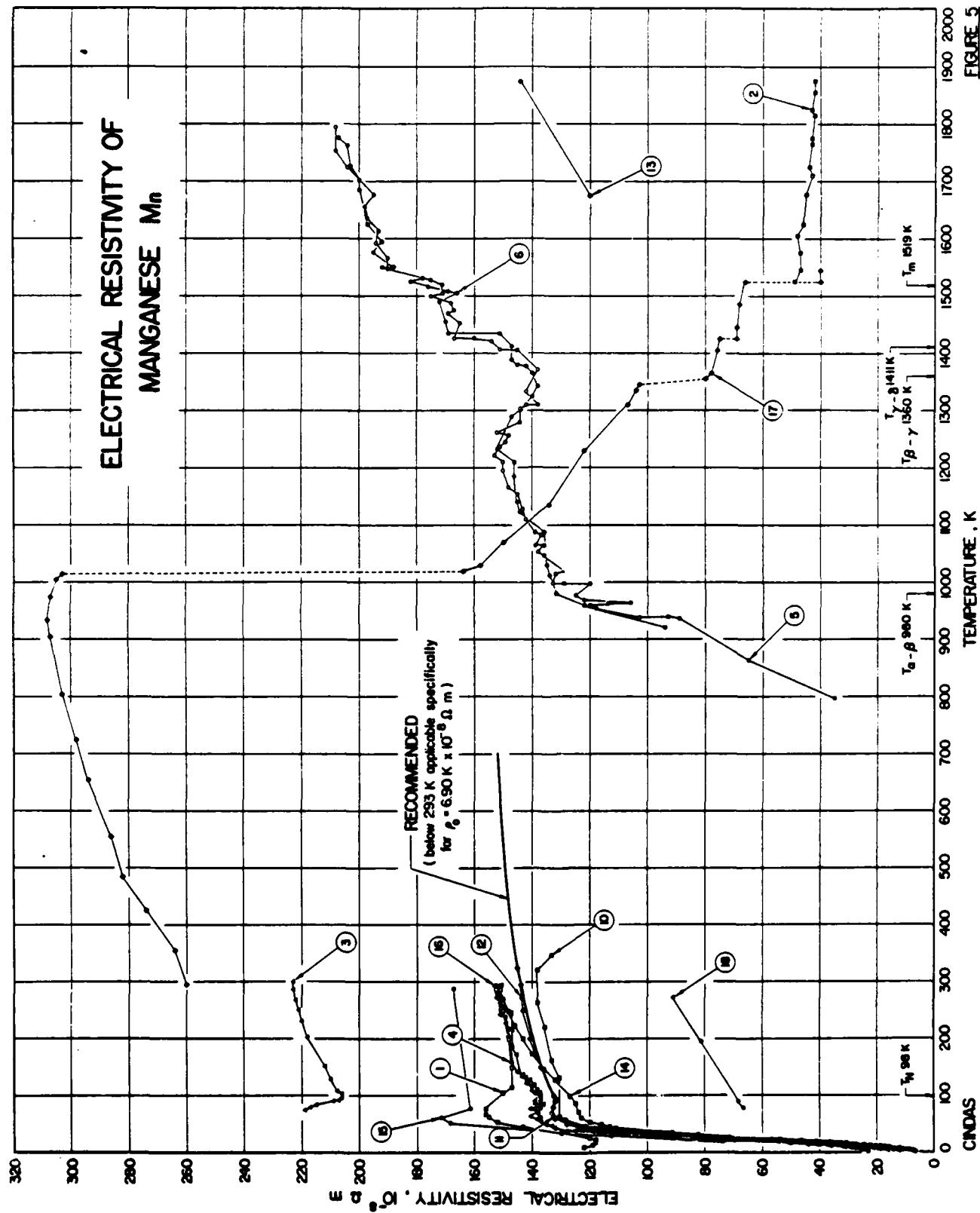


TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF MANGANESE - Mn

Data Ref. Set. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1 29:	Adamu, K.G. and Grassie, A.D.C.	1978	P	10-287	Sxx(s)	99.98 Mn; electrolytic flakes from Koch Light Laboratories; cleaned in 5% HCl in methanol to remove surface oxidation and contamination; dried and ground immediately before being loaded into a previously cleaned molybdenum boat; films were prepared by thermal evaporation of Mn powder onto thin glass substrates, cut to size and cleaned; substrates heated to 473 K during evacuation of chamber then cooled to 363 K, temperature at which evaporation was carried out; coating pressure was about 10^{-6} torr; films were allowed to cool to room temperature before removing from vacuum chamber; thickness of film is 4000 Å; data read from figure; very large value of about 120×10^{-8} Ω m is attributed to several atoms driven into spin fluctuations.
2 298	Levin, E.S., Zamaryev, V.N., and Gal'd, P.V.	1976	+	1523-1873	Liquid manganese; remelted electrolytic distilled in a vacuum; average of heating and cooling experiments; measurements with contact-lens method in a revolving magnetic field; torsional oscillating method; measurements error did not exceed 7%.	
3 299	Butylenko, A.K. and Kobashev, N.S.	1976	A	75-301	α-Mn	99.9 Mn; data extracted from figure; two coordinate potentiometer.
4 300	Magashev, N. and Sembra, M.	1975	V	3-282	α-Mn	99.99 Mn; flakes were etched in HNO ₃ to remove surface oxidation; accuracy of resistance measurements is about 0.05%; uncertainty of about 10% assigned to resistivity values because of the uncertainty in determining cross-sectional area of sample; current reversed to eliminate thermal emf; data extracted from graph.
5 301	Akhavetsov, Yu.M., Baum, B.A., and Gal'd, P.V.	1969	R	797-1793	99.99 Mn; triple vacuum melted; measurements in helium using aluminum oxide crucible with closely fitted lids of the same material; resistivity of Mn increased by 5% during melting; data extracted from figure for heating experiment.	
6 301	Akhavetsov, Yu.M., et al.	1969	R	921-1775	Same as above except data for cooling experiment.	
7* 302	Meaden, G.T. and Pellow-Cervais, P.	1967	A	1.87-300	99.995 Mn; electrolytic Mn from Koch Light Laboratories; 20 ppm Mg, 2 ppm Si, <1 ppm Cu; irregularly shaped flakes of uniform thickness of about 1 mm; platelet samples were shaped by spark erosion into rectangular parallelopipeds 5-6 mm x 20-30 mm; the values in the parenthesis are for specimen after being etched in dilute HCl and annealed in vacuum 1-8 x 10 ⁻⁶ torr for 7 hr at 898 K.	
8* 302	Meaden, G.T. and Pellow-Cervais, P.	1967	A	300	Similar to the above except electrolytic manganese supplied by Pechinay of unknown purity; the values given in parenthesis are for specimens after being etched in dilute HCl and annealed in vacuum 1-8 x 10 ⁻⁶ torr for 7 hr at 898 K.	
9* 302	Meaden, G.T. and Pellow-Cervais, P.	1967	A	4.2,300	Similar to the above except electrolytic manganese supplied by Johnson-Matthey of unknown purity; the values given in parenthesis are for specimen after being etched in dilute HCl and annealed in vacuum 1-8 x 10 ⁻⁶ torr for 7 hr at 898 K.	

^{*}Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF MANGANESE—Mn (continued)

Date Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent). Specifications and Remarks	
10 303	Madden, G.T.	1966	A	1.9-348	99.995 Mn supplied by Koch Light Laboratories Ltd.; impurities such as 20 ppm Mg, 2 ppm Si, 1 ppm Cu; surface contamination was removed by reduction in dilute NaCl; annealed in 10^{-6} torr vacuum for 7 hr at 898 K; extrapolated ρ from 2 K is $6.87 \times 10^{-8} \Omega \text{ m}$; values read from figure which do not agree with some values given in text.		
11 303	Madden, G.T.	1966	A	16-70	Same as above except data extracted from table (text).		
12 304	Madden, G.T. and Peltown-Cervais, P.	1965	A	0-325	99.995 Mn; 20 ppm Mg, 2 ppm Si, <1 ppm Cu; the electrolytically made specimen was supplied by Koch Light Laboratories Ltd.; the specimen dimension $0.965 \times 4.92 \times 24.95 \text{ mm}$; the specimens were annealed under a vacuum of 10^{-6} to 8×10^{-6} torr for 7 hr at 898 K; the resistivity at 0 K was obtained by extrapolating from 2 K; error associated with resistivity data did not exceed 1%; above 80 K average of heating and cooling experiments.		
13 305	Vostryakov, A.A., Vatolina, N.A., and Zalin, O.A.	1964	-	1673-1873	Electrolytic manganese.		
14 306	Bellamy, R.V. and Coles, B.R.	1963	A	2-293	Two specimens 99.95 Mn taken from different batches of Johnson Matthey electrolytic manganese; vacuum annealed near 873 K after cutting into suitable shapes ($14 \text{ cm} \times 1 \text{ mm} \times 1 \text{ mm}$) with an ultrasonic cutter; measured resistance was converted to resistivity by assuming $\rho_{\text{at } 0^\circ\text{C}} = 130 \times 10^{-8} \Omega \text{ m}$ for pure manganese; observed Néel temperature is 75 ± 2 K; data extracted from the graphically smooth values of the authors.		
15 307	White, C.K. and Woods, S.B.	1957	A	2-288	Nr.3	Specimen from Ms. Johnson Matthey and Mallory Ltd. (JM 19792); high purity specimen with 10 ppm of Mg as major solid impurity; annealed specimen; data calculated from ρ_1 values represented graphically using $\rho_0 = 11.3 \times 10^{-8} \Omega \text{ m}$ reported by authors.	
16 307	White, C.K. and Woods, S.B.	1957	A	6-295	Mn	Specimen cut from material supplied by A. D. Mackay Inc.; annealed in vacuum at 873 K for some hours to remove adsorbed hydrogen; spectrographic analysis showed that this material was of comparable high purity to that of Mn; data extracted from figure; data exhibits a shallow minimum near 100 K and falls rapidly below 50 K; residual resistivity $\rho = 16.8 \times 10^{-8} \Omega \text{ m}$.	
17 308	Grube, G. and Speidel, W.	1940	R	293-1543	Vacuum distilled Mn; 0.01-0.001% Fe and Si; <0.001% of Cu, Ca, and Al; cylindrical specimen 9 mm diam. and 15 mm length.		
18 309	Redemann, H.	1935	-	78-273	β -Mn	No details given except sample ~16 mm long and ~5 mm diameter and end surfaces were ground; values calculated from reported $\rho_{127^\circ\text{C}}$ values and $91.0 \times 10^{-8} \Omega \text{ m}$ for electrical resistivity at 273 K.	

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF MANGANESE Mn (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
19 ^a	310	Murayama, S. and Nagasawa, H.	1977	A	1.17-4.15	0-Mn	Pure Mn; specimen same as the one reported in data set 4; annealed at 625°C for 48 h to obtain pure 0-Mn and at 600°C for 24 h to remove strain during sample measurements in 0 kOe; longitudinal and transverse magnetoresistance.

^aNot shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF MANGANESE
[Temperature, T, K; Electrical Resistivity, ρ , $10^{-9} \Omega \cdot m$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ			
<u>DATA SET 1</u>																		
10.6	122	96.7	206.1	98.9	138.0	106.3	139.1	92.1	94	4.2	(13.7)							
14.4	119	102.6	206.1	105.7	139.7	108.0	137.9	95.8	122	4.2	(11.2)							
18.2	118	108.4	207.9	109.1	140.5	108.6	139.7	97.8	132	4.2	(9.1)							
22.0	118	129.4	209.8	117.5	140.5	110.9	142.1	101.0	134	1.87	(7.3)							
26.8	120	151.6	212.2	122.6	142.1	112.7	143.3	102.8	135									
35.2	130	203.0	217.7	129.3	142.1	113.3	145.1	108.6	136									
44.7	143	231.0	219.5	132.7	143.0	116.5	148.6	112.1	144									
54.1	152	250.9	221.4	137.8	143.8	119.4	150.4	113.9	145	300	345(185)							
61.7	155	268.4	222.0	142.8	145.5	120.9	150.4	118.3	146	300	320(175)							
67.4	156	287.1	222.6	169.8	147.1	122.0	153.9	120.9	146									
71.2	156	301.1	222.6	183.3	147.1	123.5	151.0	123.2	152									
76.8	156	190.0	190.0	148.0	124.4	149.3	128.8	147										
105.2	150	<u>DATA SET 4 (cont.)</u>																
114.7	147	218.7	148.7	125.6	168.1	132.3	140	140										
120.3	147	3.5	16.8	228.8	149.5	127.9	144.6	134.1	138									
135.5	147	5.2	19.4	240.6	149.5	129.1	144.7	136.4	140									
150.6	147	6.9	21.0	24.0	151.2	130.3	146.7	137.6	142									
197.9	148	8.6	26.1	250.7	150.3	130.9	142.4	137.9	145									
250.8	150	8.7	31.2	257.5	150.3	130.9	138.6	138.8	147									
286.8	151	10.4	32.8	264.2	152.0	133.2	142.4	141.1	147									
		10.4	36.2	282.8	152.8	137.0	138.9	143.4	151									
<u>DATA SET 2</u>																		
152.3	49	13.8	41.3	<u>DATA SET 5</u>														
154.3	47	17.3	51.4	44.6	13.9	46.5	140.5	151.3	1454	170								
157.3	47	20.7	60.6	79.7	35.9	106.8	142.5	160.0	150.7	169	5.74	10.33						
160.3	48	20.8	70.7	86.3	65.2	114.4	142.5	167.0	148.9	172	6.03	10.68						
162.3	46	20.9	77.5	93.5	89.9	145.2	165.9	151.6	176	1516	6.50	11.22						
167.3	45	24.4	89.3	93.8	93.9	146.9	169.4	153.0	178	178	6.84	11.94						
170.8	43	27.8	92.6	96.4	106.8	148.7	167.7	152.4	162	172	7.56	12.55						
172.3	44	33.0	109.5	96.1	114.4	150.4	171.2	155.0	192	1550	188	9.25	15.83					
176.3	43	36.4	117.1	95.2	118.5	151.9	171.8	157.6	195	172	10.2	17.31						
177.3	43	39.8	125.5	95.8	120.3	149.8	175.3	159.4	192	176	11.6	20.23						
181.3	42	39.9	128.0	97.5	119.1	152.7	175.9	164.4	197	178	12.2	22.23						
182.3	43	44.9	132.2	96.7	122.0	154.7	170.6	168.5	200	168.5	182	13.8	24.31					
185.3	42	48.3	133.9	97.8	121.4	156.5	180.6	170.2	200	170.2	188	14.5	28.37					
187.3	42	50.0	135.6	99.6	120.9	159.1	194.7	172.6	203	172.6	192	15.3	30.51					
		56.6	137.2	97.5	125.5	161.2	191.6	176.1	204	176.1	192	16.1	32.87					
<u>DATA SET 3</u>																		
75.6	218.8	73.7	138.0	99.6	133.2	165.5	198.9	188.2	200	188.2	200	18.2	46.38					
80.3	217.0	80.4	137.2	101.3	132.6	172.3	204.8	195.4	204	195.4	204	22.4	53.75					
81.8	214.6	83.8	137.2	104.5	136.1	175.2	208.4	196.7	203	196.7	203	22.9	56.50					
89.7	209.1	87.1	136.3	105.1	138.5	179.3	208.4	197.7	207	197.7	207	16.1	32.87					
92.0	207.3	92.2	137.2	106.3	136.2	172.2	205(144.2)	198.9	205(144.2)	198.9	205(144.2)	31.9	82.29					
<u>DATA SET 7 (cont.)*</u>																		

*Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF MANGANESE Mn (Continued)

Not shown in figure.

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5. APPENDICES

5.1. Methods for the Measurement of Electrical Resistivity

At the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) of Purdue University, the experimental methods for the measurement of electrical resistivity have been classified into various categories according to a similar scheme used by CINDAS for the classification of methods for the measurement of thermal conductivity [356, pp. 13a-25a]. This classification scheme of CINDAS is presented below. Note that the letters in parentheses following the respective methods are the code letter used in the 'Method Used' column of the Table of Measurement Information for indicating the experimental methods used by the various authors.

Methods for the Measurement of Electrical Resistivity

A. Steady-State Methods

1. Voltmeter and ammeter direct reading method (V) [357, p. 159; 358, pp. 244-5]
2. Direct-current potentiometer method (A) [359, pp. 151-8]
 - a. 4-probe potentiometer method
3. Direct-current bridge methods (B) [359, pp. 144-51]
 - a. Kelvin double bridge method
 - b. Mueller bridge method
 - c. Wheatstone bridge method
4. Van der Pauw method (P) [360, 361]
5. Direct heating method (K) [362, 363]

B. Non-Steady-State Methods

1. Periodic current method
 - a. Direct connection to sample
 - (1) Alternating-current potentiometer method (C) [359, pp. 161-2]
 - (2) Alternating-current bridge method (D), [359, p. 162]
 - b. No connection to sample
 - (1) Rotating magnetic field method (R) [364]

5.2. Conversion Factors for the Units of Electrical Resistivity

The recommended values and experimental data for the electrical resistivity tabulated in this work are in the units: $10^{-8} \Omega \text{ m}$. Conversion factors for the units of electrical resistivity, which may be used to convert the values given in $(10^{-8} \Omega \text{ m})$ to values in other units, are given below.

Conversion Factors for the Units of Electrical Resistivity

Units to be Converted to	Multiply the Value Given in $(10^{-8} \Omega \text{ m})$ by
ohm-meter ($\Omega \text{ m}$)	1×10^{-8}
ohm-centimeter ($\Omega \text{ cm}$)	1×10^{-6}
ohm-inch ($\Omega \text{ in.}$)	3.937×10^{-7}
ohm-foot ($\Omega \text{ ft}$)	3.281×10^{-8}
microohm-centimeter ($\mu\Omega \text{ cm}$)	1
abohm-centimeter ($ab\Omega \text{ cm}$)	1×10^3
statohm-centimeter ($stat\Omega \text{ cm}$)	1.113×10^{-18}
emu (= ab Ω cm)	1×10^3
esu (= stat Ω cm)	1.113×10^{-18}
ohm-circular mil per foot ($\Omega \text{ cmil ft}^{-1}$)	6.015

Example: $1.000 \times 10^{-8} \Omega \text{ m} = 3.937 \times 10^{-7} \Omega \text{ in.}$.

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